## **UNCLASSIFIED**

# AD NUMBER AD859274 **NEW LIMITATION CHANGE** TO Approved for public release, distribution unlimited **FROM** Distribution authorized to U.S. Gov't. agencies only; Administrative/Operational Use; NOV 1968. Other requests shall be referred to US Army Material Command, Alexandria, VA. **AUTHORITY** WES, per d/a ltr dtd 6 Oct 1977

# AD 859274

Contract Report S-68-5

### Research Study on

SOIL TREATMENT MATERIALS FOR
DUST PALLIATION, SOIL WATERPROOFING AND
SOIL STRENGTHENING

by

C. N. Impola and D. A. Olsen

NOVEMBER 1968

### Sponsored by

U. S. ARMY MATERIEL COMMAND PROJECT NO. 1-T-0-62103-A-046-05

### Conducted for

U. S. Army Engineer Waterways Experiment Station CORPS OF ENGINEERS Vicksburg, Mississippi

Under

CONTRACT NO. DA-22-079-eng-437

bу

ASHLAND CHEMICAL COMPANY Research Department 10701 Lyndale Avenue South Mirneapolis, Minnesota 55420

ARMY-MAC VICKEBURG, MISS.

SEP 25 1969

Each transmittal of this document outside the agencies of the U. S. Government must have prior approval of U. S. Army Materiel Command.

### TABLE OF CONTENTS

FOREWORD	Page 1
PART I: BACKGROUND	
Introduction	5 5
PART II: MATERIALS TESTED AND TEST PROCEDURES	
Materials	
Soils	4
Resins	4
Sample Preparation	5
Tests Performed	6
PART III: TEST RESULTS	
Preliminary Screening of Resins	9
Physical Testing	- 1
Poured or Admixed Applications	14 17
Flexural Strength of Sand with Unsaturated	<b>.</b> .,
Polyester Resins	19
Storage Stability of Unsaturated Polyester Resins	55 50
Orecimine Elascomers	22
PART IV: PHYSICAL CHEMISTRY STUDY	
Estimation of Surface Free Energies	26
Contact Angle Measurement	26
Determination of Surface Free Energy	28
Correlation of Spreading Coefficients with Compressive	
Strengths	29
PART V: CONCLUSIONS AND RECOMMENDATIONS	
Conclusions	32
Recommendations	34
FIGURES 1-6	
_	

TABLES 1-8

### FOREWORD

The work in this report was performed under Contract Number DA-22-079-eng-437 entitled "Research Study on Soil Treatment Materials for Dust Palliation, Soil Waterproofing and Soil Strengthening," dated June 15. 1965, between United States Army Waterways Experiment Station and the Ashland Chemical Company. This research was sponsored by the United States Army Materiel Command under Department of the Army Project DA-1-T-0-21701-A-046, "Trafficability and Mobility Research," Task 05, "Mobility Engineering Support (Dust Control, Southeast Asia)."

This report was prepared by Mr. C. N. Impola and Dr. D. A. Olsen of Ashland Chemical Company. This program was under the general supervision of Dr. W. J. McKillip of the Contract Research Group of Ashland Chemical Company.

The contract was monitored by Mr. G. R. Kozan, Chief, Stabilization Section, Expedient Surfaces Branch, under the general supervision of Mr. W. J. Turnbull, Chief, Soils Division, WES. Contracting Officers were COL J. R. Oswalt, Jr., CE, and COL Levi A. Brown, CE.

### PART I: BACKGROUND

### Introduction

Military operations require improved materials to suppress dust at airfields, helipads, operational bases, intermediate and cantonment areas as well as supply bases and roads. Dust in these areas has increased logistical problems greatly by reducing service life of machinery and equipment. For example, helicopter rotor blades have to be replaced after 200-300 hours rather than the expected 1100 hours. Also, the engines have to be replaced after 1/3 to 1/2 the normal usage period.

Present requirements for a potential dust palliative have been set forth in the Department of the Army Approved Qualitative Materiel Requirement (QMR) for Dust Control Material (August 1966). The following performance characteristics are sought:

- 1. Be effective and operationally usable within four hours after application to the surface of all types of soil.
- 2. Withstand, without failure or peeling, helicopter rotor downwash (10 psf disc loading) and C-130 aircraft propwash (100 mph air velocity).
- 3. Be effective, with only minor maintenance, for the following minimum time:
  - a. Six months in non-traffic areas.
  - b. Three months in areas subjected to infrequent traffic of ground vehicles or aircraft, such as shoulders or overruns of runways.
  - One month in areas trafficked by ground vehicles or aircraft.

In addition to the above performance requirements, the following usage requirement is requested:

Weight and volume characteristics of the material shall not exceed three pounds per square yard or 0.45 gallons per square yard of ground surface treated on trafficked areas.

### Purpose and Scope

Initial screening of Ashland's resins for use as dust palliatives in non-trafficked areas as authorized under Contract Number DA-22-079-eng-437 was based on knowledge, technology and resins

developed for foundry core binders. The determination of surface free energies of soils, and correlation of wetting with compressive strengths of treated soils was also authorized under this contract.

Laboratory modifications of these resins and screening of other Ashland resins along with other commercially available organic and inorganic resin systems were done under Modification Number 1 of the subject contract.

Under Modification Number 2 of the subject contract, laboratory synthesis and evaluation of new and specialized resin systems were conducted to meet the broader requirements of the QMR which was issued during this period.

This final report, prepared in fulfillment of Contract Number DA-22-079-eng-437 and Modifications Number 1 and 2, covers the work done from June 15, 1965 to January 22, 1967.

### PART II: MATERIALS TESTED AND TEST PROCEDURES

### Materials

### Soils

Three types of soil procured locally and 150 pounds of buckshot clay received from Waterways Experiment Station (WES), Vicksburg, Mississippi, were used in the screening of the resins as dust palliatives. The soils procured locally were dried and pulverized at 150 F. using a drum drier in the Ashland experimental station. Final water content was less than 2%. The sieve analyses obtained on these soil samples by the Ashland Control Laboratory are given below.

Per	Cent	Passing
LET	Cent	LGPSTUS

Screen Size No.	Sand	Clay	Silt
10	99.I		100
20		92.4	93.2
40	88.5	72.2	72.9
60	65.1	48.2	53.7
8 <b>o</b>		51.4	
100	30.7	47.4	35.9
200	10.4	<b>3</b> 8.5	27.0

The buckshot clay received from WES was not analyzed for size distribution.

### Resins

The resins that were evaluated as dust palliatives were chosen on the basis of:

- 1. Experience with foundry core resins
- 2. Known resin properties such as ease of cure and physical strength
- 3. Ease of app ::ation
- 4. Availability
- 5. Cost

The following general classes of resin systems were evaluated:

Epoxy Emulsified petroleum resin Gelatin Acrylamide Acrylic Phenol-formaldehyde Analine-furfurol Polyurethane Sodium silicate N Plasticized sulfur Unsaturated polyester Alkyd

### Test Procedures

### Sample Preparation:

Primary evaluations were conducted by placing the loose dried soil into an 8-inch diameter by 1 1/2 inch deep disposable pie tim. The resin was then poured on the surface. Limited supplies of dried and pulverized soils required changing to 2-inch diameter sample size for initial screening. Samples were prepared using 2-inch ID rigid polyvinyl chloride pipe cut in 2 1/2-inch lengths. The test samples were prepared in three different ways:

- 1. Two inches of loose soil was placed in the mold and the calculated amount of resin to give 3 pounds per square yard was poured on and allowed to soak into the soil.
- 2. Soil and resin were prepared as in 1 above, but the samples were compacted after one hour using an American Foundry Service Rammer made by the Dietert Company as specified for foundry sample preparation. Compaction was effected by dropping a 14-pound weight attached to a two-inch diameter disk from a height of two inches three times to the 2-inch diameter surface.
- Soil and resin thoroughly mulled or ad ited by mechanical stirring and then compacted as in 2 above.

Other test samples were prepared by mulling or admixing the soil and resin by mechanical stirring and then cast into a  $4\times8\times1/2$ -inch mold, pressing out the entrapped air with a trowel and the surface troweled smooth.

Another method, used for spray application, was to pour the loose, dry soil into a  $6 \times 6 \times 2$ -inch mold. The soil was leveled to the mold edge with a straight edge, then the resin was applied with either air atomizer or air-less spray gun.

To obtain physical properties on urethane elastomers, the elastomers were cast on a round one-gallon can cover treated with wax to prevent adhesion. The liquid urethane was poured on the cover and allowed to cure. After cure, the free film was removed from the cover and dumbbell-shaped specimens were cut for testing using a die according to ASTM Method  $D_{-}^{h}12$ .

### Tests Performed

The following tests were run on samples which had no visual defects on the surface, such as cracks:

- 1. Six by six-inch diameter soil specimens
  - a. Effect of air impingement was evaluated by subjecting the specimens to an air blast equivalent to 150 mph for one minute. Specimens were tilted at an angle of 10° in the direction of the air clast. The air blast was created by a blower with a rated capacity of 8000 cfm.
  - b. Samples that passed the air impingement test, i.e., where the surface skin was not broken, were checked for resistance to water erosion. The test specimens were mounted below a spray nozzle calibrated to give a 65° solid water come at a rate of 1.1 gallons per minute at 40 psi water pressure. The samples were placed 4-1/2 inches below the nozzle tips so that 70% of the sample was sprayed with high velocity water droplets. The duration of the test was two hours initially. This was later reduced to one hour duration to conform to the tests run at Waterways Experiment Station, Vicksburg, Mississippi. Weight immediately after testing was recorded and compared to weight before testing to determine if any water had been absorbed through the resin surface.
  - c. Samples that passed the water spray test were again subjected to the air impingement test as in "b" above.
- 2. Two-inch diameter and 2 x 2 x 1 1/2-inch soil specimens
  - a. Water repellancy tests were run on the 2-inch diameter soil specimens by adding 0.5 ml water to the soil surface two hours after the resin application. The time required for the water to soak in was recorded. The test specimens were rated as follows:

- Poor (P) water soaked in within one minute; Fair (F) water visible on the surface 1-5 minutes: Good (G) water visible on the surface 5-10 minutes; and Excellent (E) water visible on the surface more than ten minutes.
- b. Unconfined compressive strengths were run on fully cured (at least three days at ambient temperature) 2-inch diameter specimens and 2 x 2 x 1 1/2-inch samples cut from the 8-inch diameter specimens. Surfaces of the specimens were ground smooth with a belt sander before testing. Compressive strengths were run using a Model TT-C Instron Universal Testing Machine according to ASTM Method D-2166 with a head speed of 0.05 inches per minute.
- c. Freeze-thaw resistance was evaluated by immersing the specimens for 16 hours in water at room temperature, pouring off the water and placing the wet specimens in a freezer for 8 hours at -10 F. Appearance and unconfined compressive strengths were recorded after 8 cycles.
- d. Wet-dry resistance was determined by immersing the 2-inch diameter or 2 x 2 x 1/2-inch specimens cut from larger samples for 8 hours in water at room temperature, pouring off the water, and subjecting the specimens to heat for 16 hours in a forced draft oven at 140 F. The specimens were tested for unconfined compressive strength after 8 cycles.
- 3. Four-by-eight by one-half inch cast specimens
  - a. Water repellancy tests were run on the cast soil specimens after two hour cure according to the procedure in 2a.
  - b. Flexural strengths were determined by the standard ASTM Method D709 using a Model TT-C Instron Universal Testing Machine. Three  $1\times8\times1/2$  inch specimens were cut from the cast  $4\times8\times1/2$  inch specimens using an abrasive blade.

### 4. Cast urethane elastomers

a. Tensile strengths and per cent elongations were determined for the elastomers by ASTM Method D-412 using a Model TT-C Instron Universal Testing Machine at a head speed of 20 inches per minute.

Some of those elastomers having elongation beyond the range of the Instron were tested on a Scott Model L-6 Tensile Tester with a head speed of 20 inches per minute.

### PART III: TEST RESULTS

### Preliminary Screening of Resins

Sixteen different resin systems, including organic and inorganic, were evaluated as dust palliatives and soil waterproofing agents in the initial screening. Each resin type is discussed below in the order of listing in table 1.

### 1. Bisphenol A Derived Epoxy Resins

Bisphenol A type epoxy resins are used where good adhesion and high tensile strengths are desired. Epon 828 from Shell Chemical Company was selected as representative of this group of resins. A low-cost, amine-terminated resin, Ashland Serial 1496, was selected as the curing agent because of its low toxicity, ready availability, and Ashland's prior experience with this material. Good penetration was obtained on sand and clay when the resin was diluted with high levels of Solox solvent. The cure times were slow (12-40 hours). The clay-epoxy specimen broke up easily.

### 2. Emulsified Petroleum Resins

Coherex, an emulsified petroleum resin manufactured by the Golden Bear Oil Company, has been widely used as a dust palliative around military installations. Coherex and a comparable Ashland experimental emulsion were evaluated on sand and clay. The penetration was good on sand but only minor penetration was obtained on clay. The treated surface did not form any skin and stayed moist. The Coherextreated special and only fair water repellency after two-hour cure.

### 3. Gelatin

15XPF gelatin was tested on dry sand with chromium sulfate, and formaldehyde as catalysts. The sample treated with chromium sulfate apparently gelled, preventing complete penetration. The formaldehyde-cured sample gave complete penetration in the sand and cured to a hard, but easily broken, brittle composite. The chromium sulfate-cured specimen was softer and rubbery.

### 4- AM-9

AM-9, a chemical grout marketed by American Cyanamid, is a blend of water-soluble acrylamide and diacrylamide

which polymerizes when properly catalyzed to give a voidfilling substance for sealing leaks in dams, etc. When catalyzed with DMAPN-KFe\*, AM-9 rapidly cures to a crumbly, soft gel that continues to cure over a period of several days to a hard, strong composite. Water repellency of the treated sand specimen was poor.

### 5. Acrylic Emulsions

Acrylic emulsions are relatively low-cost, water-dilutable resin systems and are widely used as paint vehicles, adhesives. fabric treatments, etc. Based on recommendations of the Ashland Resins Laboratory, a number of experimental emulsions were screened. EP8908-23, EP8908-122 and EP8908-129 are representative of acrylic emulsion resins that may be obtained commercially.

The resins as supplied, had a viscosity too high to penetrate the soil samples to any degree. However, they formed a heavy impervious surface skin on the clay and silt. The sand-treated specimen also had a surface skin; but the skin was slightly porous, testing only fair on water repellency.

Reducing the viscosity of the emulsions with water gave good penetration in sand and improved the penetration slightly in the silt and clay. However, the surface skin was more porous with the higher water content resins and the water repellency diminished considerably.

### 6. Chem-Rez 200

Chem-Rez 200 and phenol-modified Chem-Rez 200 are furfural based, rapid-setting resins developed by Ashland for foundry applications. These resins, admixed with foundry sand (3% level), give significantly higher strengths than aniline-furfural in foundry cores. When these resins were applied on sand and cured with phosphoric acid, they cured very slowly or not at all. Water repellancy was generally poor.

### 7. Aniline-Furfural

Attempts to use aniline-furfural resins as stabilizers for surface soil using either poured or admixing techniques were not very successful. Resin penetration was excellent but cures with 37% HCl required 12 to 24 hours. Erratic

<sup>\*</sup>Dimethylamino-propionitrile-potassium ferricyanide-ammonium persulfate.

cures were obtained, particularly with clay and silt soils.

### 8. Emlon E-200

Emlon E-200 is a new epoxy resin manufactured by Stoner-Mudge, Pittsburgh, Pennsylvania, that is soluble in water. Cures of the Emlon E-200 were obtained in two hours or less with diethylene triamine, which is also water soluble. Reproducibility of cure and composite homogenity appear to be problems based on our initial screening.

### 9. Polyurethane Elastomer

An attempt was made to use the dried sand as a filler for a urethane elastomer-type composite. Sand was blended with a 2000 molecular weight polypropylene glycol (Wyandotte P2010) and then with a low-cost crude disocyanate (Allied Nacconate 4040). Cure was very slow even with substantial quantities of stannous octoate, and residual moisture in the sand caused extensive foaming of the composite.

### 10. Sodium Silicate N

Penetration of sodium hexafluorosilicate-cured sodium silicate N was low by the poured technique. Composites prepared by admixing cured in one hour to a hard, somewhat brittle composite. The water repellency and surface abrasion of the sodium silicate N sand specimens were poor.

### 11. Sulfur

Sulfur plasticized with E-13, a thioglycollate ester, gave hard, chemically-inert composites on cooling from the molten state (approximately 150 C) to room temperature. The plasticized sulfur and soils were preheated to 150 C, blends were prepared with sand and silt, and then the blends were poured into molds. The composites set to rigid solids on cooling, generally within ten minutes.

### 12. Unsaturated Polyester Resins

The low cost of polyesters (less than  $25\phi/lb$  in bulk) and the ability to use low-cost reactive solvent (styrene - less than  $10\phi/lb$  in bulk) prompted a rather extensive evaluation of these resins for soil treatment. Complete penetration of all three soil types (sand, clay, and silt) was readily obtained when high levels of styrene (up to  $80\phi$ ) were used to reduce resin viscosity to essentially the consistency of water. The cure times could readily be controlled over

wide limits by changing catalyst and/or promoter levels. Addition of 10% water to the soil had no observable effect on gel time. Excellent water repellency was obtained when higher resin solids unsaturated polyester resin was used.

### 13. Moisture-Cure Polyurethanes

Polyurethane resins that cure by reaction with moisture in the air are rapidly gaining in popularity as coatings and sealants. The urethane resins can be readily formulated to give plastics that are hard and brittle or strong elastomeric rubbers.

Initial results using Arothane 160, Arothane 156 and Arothane 170 as poured resins for soil stabilization were very encouraging. These resins, particularly Arothane 170 diluted with ethyl acetate to reduce viscosity, gave good cures as soon as one hour if properly catalyzed. These resins were outstanding in their ability to waterproof all three types of soils.

### 14. Water-Soluble Alkyds

Alkyd resins offer the unique advantage of reaction with oxygen in the air to give surface cures for dust palliation and soil waterproofing. The resin below the surface should remain in an uncured state to provide self-healing properties if the surface skin is broken. Arlon 110, Arlon 310, Arlon 363, and Arlon 580 were selected as representative resins for the early screening.

Penetration of these resins was unexpectedly poor. The resins tended to give the desired surface skin, but the surface strengths were poorer than other systems tested. Water repellency was generally only fair, presumably due to the presence of water solubilizing groups in the resin.

### 15. Solvent-Dilutable Alkyds

The solvent-dilutable alkyds would also be expected to give only surface cure for use as dust palliatives and waterproofing agents. Ashland has pioneered the development of rapid-setting alkyd resins, which permit the formulation of solvent resistant enamels that have short drying times. Aroplaz 6008, Aroplaz 6065, and Aroplaz 832 were selected as representative resins for this study.

As expected, these resins gave surface skins when applied by poured technique. These films were excellent water repellence for sand and clay. Penetration into sand was excellent. Only slight penetration of clay was observed.

### 16. Miscellaneous Systems

### a. Aroflint 505

Aroflint 505 is a specially-epoxidized oil resin that gives hard, ceramic-like coatings on cement blocks, etc. This resin did not cure when applied to soil samples and water repellency was poor as the water soaked into the soil immediately.

- b. Several newly-developed foundry resins (Lino-Cure, Lino-Cure A, Lino-Cure C, Lino-Cure 2125) were evaluated as soil treatment materials. Lino-Cure, Lino-Cure A and Lino-Cure 2125 did not cure when applied on sand containing 10% water. Lino-Cure C applied alone did not cure either but when Lino-Cure C was applied with ethylene glycol, a hard surface cure was obtained that had excellent water repellency.
- c. Octadecyl isocyanate was evaluated as a noncuring soil waterproofing agent on sand and clay containing 10% water. Waterproofing of the soils was outstanding. Grinding the treated soils with a mortar and pestle did not reduce the water repellency. Water poured on the soil after the treatment remained in a spherical droplet. Blending the isocyanate with Arothane 170 gave soil treatment with excellent water repellency with some degree of cure to improve dust palliation properties.
- d. An unsaturated polyester resin prepared in the laboratory, EP2379-60, blended with a small amount of octadecyl-isocyanate produced excellent water repellency when applied on sand. The cure, however, was slow and incomplete, producing a rubbery skin on the surface.
- e. Several alternate polyurethane and free diisocyanate systems were examined briefly. Alpha-methyl glucoside polyether-toluene diisocyanate (AMG-TDI) prepolymers gave only limited penetration. The water repellency was good on sand and excellent on loose clay. A good abrasion resistant surface was obtained.

Hexamethylenediisocyanate alone applied on sand and clay containing 10% water gave hard composites but water repellency was poor. Most of the isocyanates are lachrymators when in the free state and not reacted on to the polymer chain.

f. Siroc #l is an inorganic siliceous grout from Diamond Alkali Company. Penetration on sand and clay containing 10% water was good. The samples did not cure and water repellency was poor.

### Physical Testing

### Poured or Admixed Applications

Table 2 lists the results of air impingement and water erosion tests, unconfined compressive strengths, and freeze-thaw and wet-dry cycling effects on the resins initially screened as soil treating materials. Each type of resin will be discussed in the order listed in the table.

### 1. Bisphenol A Type Epoxy Resins

The Epon 828-treated sand and silt specimens had good wind and water erosion resistance at the use levels tested. The lowest use level tested was 1.3 lbs/square yard resin plus 5.1 lbs/square yard solvent. Unconfined compressive strength was 1079 psi at a use level of 5.2 lbs/square yard resin and 5.1 lbs/square yard solvent. Using a level of 11 lbs/square yard resin plus 16.5 lbs/square yard solvent, the unconfined compressive strength was over 2200 psi. The Epon 828 treated sand specimens withstood freeze-thaw and wet-dry cycling without noticeable change. At lower levels, only surface penetration was obtained and a hard crust formed.

### 2. Emulsified Petroleum Resins

Coherex and Ashland experimental emulsion both withstood 60 mph air impingement test on sand (150 mph air impingement test was not used at this time). Both resin systems failed on loose clay and silt due to shrinkage cracks on the surface. Neither resin passed the water erosion test. Unconfined compressive strengths were not run since the specimens did not cure.

### 3. Gelatin

Sand treated with 15XPF gelatin cured with chronium

sulfate or formaldehyde passed both the 60 mph air impingement and water erosion test. However, the sample was flexible and easily broken after the water erosion test. Unconfined compressive strength was not run on these samples.

### 4. AM-9

AM-9 treated sand gave a hard, strong composite with an unconfined compressive strength of over 1700 psi. However, the cure time was long and resistance to freeze-thaw and wet-dry cycling was fair to poor as evidenced by the loss of 30 to 60% compressive strength.

### 5. Chem Rez 200

Chem Rez 200 applied to washed and dried foundry sand gave 354-610 psi unconfined compressive strength after 24 hour ambient cure based on results obtained by the Ashland Chemical Foundry Group. This resin, when applied on sand and cured with phosphoric acid under this test program, cured very slowly or not at all. Resistance to 60 mph air impingement was good, but water erosion was only fair with some surface erosion.

### 6. Aniline-Furfural

Unconfined compressive strengths on aniline-furfural treated sand was low (399 psi). Resistance to wind and water erosion, freeze-thaw, and wet-dry cycling was good to excellent. When applied on clay with a high level of solvent, a hard specimen was obtained after 24 hour cure. The unconfined compressive strength was very low (32 psi).

### 7. Emlon E-200

Emlon E-200 applied on sand and clay performed satisfactorily in the air impingement test. Some erosion of the treated soil, especially clay, is apparent from the loss in weight during the water erosion test.

The unconfined compressive strength was good (1388 psi). The treated specimen withstood the freeze-thaw cycling but split after the third wet-dry cycle.

### 8. Sodium Silicate N

Sodium silicate N sodium hexafluorosilicate system

added to sand by mulling formed a hard composite within one hour. Resistance to wind and water erosion was good to excellent. The unconfined compressive strength was fair (650 psi). Resistance to freeze-thaw and wetdry cycling was poor as indicated by the  $\cos 50\%$  of the unconfined compressive strength.

### 9. Unsaturated Polyester Resin

Aropol 7110 with high levels of styrene gave excellent unconfined compressive strengths when applied on sand by pour-on technique. The strengths varied 1173 psi with low resin level to 2300 psi with high resin level. The freeze-thaw and wet-dry cycling did not affect the unconfined compressive strengths. Resistance to air impingement and water erosion was excellent.

### 10. Moisture-Cure Polyurethane Resins

Arothane 170 in ethyl acetate gave satisfactory unconfined compressive strengths (700 psi) when applied on sand by pour-on. The resistance to freeze-thaw and wet-dry cycling was excellent.

### 11. Air-Dry Alkyds

Application of air-dry alkyds to sand by mulling gave hard but relatively weak composites. Arlons, the watersoluble alkyds, and Aroplazes, the solvent-dilutable alkyds, produced specimens with low compressive strengths (90 and 107 psi). Both specimens withstood the freeze-thaw cycling. On wet-dry cycling, the wateralkyd-treated specimen crumbled. The solvent-alkyd treated sample did withstand the wet-dry cycling although some spalling was noted on one specimen.

### 12. Acrylic Emulsions

Only one acrilic analysish sample, prepared with EP8908-23, was considered satisfactory for testing. Resistance to air and water erosion and wet-dry cycling was good. Although the unconfined compressive strength (685 psi) was 1/2 to 1/3 of the epoxy and polyester systems, it was considerably stronger than many of the other acrylic systems tested. One of the acrylic emulsion-treated samples, EP8908-122, crumbled in the freeze-thaw and wet-dry cycling tests. The third emulsion system passed the freeze-thaw but crumbled during wet-dry cycling.

### Sprayed Application

Table 3 lists the results of air impingement and water erosion tests of various resins applied on sand and/or clay by spray-on application. The polyester and Chem Rez resins were applied with an air atomizing spray. The rest of the resins were applied with an air-less spray gun. The results will be discussed as listed in table 3. These were tested as soil treatment materials intended for use in nontraffic areas at three pounds per square yard maximum use level, including solvent.

### 1. Asphalt Emulsions

Cationic asphalt emulsions at various solids content were tested on sand and clay. The sand samples passed the 150-mph air impingement test, both before and after the water erosion test. The surface did not fail during the water erosion test but small holes in the surface allowed some water seepage as indicated by the increase in weight.

When applied on loose clay, large surface cracks appeared due to wetting and subsequent shrinkage of the clay.

### 2. Epoxy Resins

Two different types of epoxy resins were tested on loose clay. Neither the Emlon E-200 nor the Epon 815 performed satisfactorily due to surface cracks from the shrinkage of the clay.

### 3. Acrylic Resin Emulsions

The acrylic emulsions also failed to pass the wind test on loose clay due to shrinkage cracks on the surface.

### 4. Polyurethane Resins

Arothane 170 failed the air impingement test on dry, loose clay due to poor wetting of the clay and crack. ing. Arothane 160 performed satisfactorily when applied on pre-wetted clay. The specimens passed the 150-mph air impingement tests and water erosion test. However, some water seeped through the surface in the water erosion test.

### 5. Air-Dry Alkyds

Aroplaz 6008, Aroplaz 6065, Aroplaz 832, and a blend of

Aroplaz 832-Aroplaz 6000 were sprayed on dry and prewetted clay and dry sand. The alkyds did not wet the dry clay sufficiently to cover the surface. However, after pre-wetting the clay, the coverage was good. The alkyds formed a surface skin on the clay that withstood the air impingement and was fairly impervious to water. When tested four hours after the application, the surface skin was not completely cured and tended to buckle slightly under the wind. The alkyds soaked into the loose sand but still formed a surface skin that held under the air impingement test.

### 6. Unsaturated Polyester Resin

Several commercial unsaturated polyester resins were evaluated as soil treatment materials. Most of the emphasis was placed on the less rigid Aropol 7510M and Aropol 7720M. Several levels of resin solids were used since the diluent, styrene, also crosslinks with the resin and is, therefore, not just a diluent. The maximum solids content that was tested was 50%, and the minimum was 30%. The polyester-treated surfaces generally passed the air impingement tests. However, in many instances, the resin pulled away from the edge of the mold, forming a small crack. Even with the crack at the edge, the surface coating did not blow away although some of the soil was blown out. The surface also withstood the water erosion test, but some water entered the soil in most cases.

Blends of Aropol 7510M and Aropol 7720M at 50% resin solids performed satisfactorily on both sand and clay. Aropol 7720M alone performed very satisfactorily on pre-wetted clay. It passed the air impingement tests and gained only 95 grams in the water erosion test. Part of the 95 grams, as much as 50 grams, was water soaked up by the wooden forms used to hold the soil.

### 7. Chem Rez 200 and Chem Rez 300

Chem Rez 200 at 50% solids performed satisfactorily on both pre-wetted and dry, loose clay and dry sand. The treated specimens passed the wind impingement tests. The specimens also passed the water erosion tests with no apparent change. The weight increase after the water erosion varied from a low of 45 grams to a high of 207 grams.

Chem Rez 300, tested at varying solids content and as blends with epoxy resins (Dow's DER, Shell's Epon) passed the air impingement tests on loose, dry clay. The weight increase during the water erosion tests was considerably lower than with other systems generally. Epon 815 blended with the Chem Rez 300 produced some flexibility to the surface coating.

### 8. Epoxidized Oils

Admex 710, epoxidized soybean oil, and ELO, epoxidized linseed oil, both performed satisfactorily on sand and clay. However, the tear strength of the surface coating from these resins appeared low.

### 9. Miscellaneous Resin Systems

Several polyurethane resins were synthesized in the laboratory to attempt to improve the flexibility of the surface coatings on soil. Series 112B to 115B in table 4 are representative of these. These resins applied on clay and sand at 50-60% solids performed satisfactorily in the air impingement and water erosion tests. However, the flexibility of these systems was less than deemed necessary.

A styrene-butadiene latex emulsion tested on sand gave a slow cure but did pass the tests. The surface had poor flexibility. The same latex applied on clay produced several cracks on curing.

EP8911-41 is a low-cost experimental latex emulsion prepared by the Ashland resin group. This material, applied on clay, gave good coverage but the flexibility of the surface coating was only moderate and the tear strength was low.

### Flexural Strength with Unsaturated Polyester Resin

Unsaturated polyester resins exhibited good physical strength properties in the early testing. A series of flexural strength specimens were prepared with Aropol 7510M mulled with sand.

The test specimens were prepared by adding a weighed amount of sand to polyester resin containing catalyst and promoters and mixing with a mechanical stirrer for 20-30 seconds. The sand mixture was then poured into a  $4\times8$  -1/2-inch mold, packed to remove entrained air, and leveled with a spatula. Three 1 x 8 x 1/2-inch test specimens were cut from each

mold using a table saw with a cartorundum blade. These cut specimens were tested for flexural strength on an Instron Model TT-C Universal Testing Hachine using Standard ASTM D-709. The experiments were designed for a statistical study by factorial analysis to determine the interaction, if any, and the optimum level, for cobalt, dimethylanaline (DMA) and hydrogen peroxide  $(H_2O_2)$ . Three levels of each component were tested on dry sand using Aropol 7510M at 30% resin solids in styrene. Cobalt and DMA were tested at 0.5, 1.0 and 1.5% of the resin-plus-styrene weight and 50% aqueous hydrogen peroxide was tested at 1.0, 1.5, and 2.0% (same basis). Data is listed in table 4.

The flexural data, obtained from these tests, were analyzed by computer. A definite interaction was found between cobalt and hydrogen peroxide. Based on these results, the optimum level of cobalt was found to be 1.35% and 1.27% the optimum hydrogen peroxide level. No clear interaction of DMA with cobalt and/or hydrogen peroxide could be determined by analysis of the data.

A second, shorter series of runs was conducted using Aropol 7510M at 30% resin solids in styrene on wet sand (10% water premixed into the sand). Three levels of cobalt, IMA, and hydrogen peroxide (50% aqueous) were used: 1, 2, and 3%. Using a simple Latin square statistical program, no conclusion as to interactions could be reached when analyzed by computer because of insufficient data. Sinificantly lower flexural strengths were obtained on wet sand as compared to dry sand (200-400 psi).

Flexural strengths were also run on the specimens after a seven-day age. (First flexural strengths were run after three days.) In general, a significant increase in strength was observed after the aging, indicating the cure continues over a period of several days.

Some of the flexural samples did not cure due to wrong proportions of promoter-to-catalyst. However, several high flexural strengths were obtained. For example, a flexural strength of 2250 psi was obtained using 0.5/0.5/1.5% ratio of cobalt, DMA and  $H_2O_2$ . Other significant values were:

Ratio - Co/DMA/H <sub>2</sub> O <sub>2</sub>	Flexural Strength psi
1/1/1%	2145
1.5/0.5/1%	2163
1.5/1.5/1.5%	2047

### Unsaturated Polyester Resins

Storage stability of unsaturated polyester-styrene resin

systems in known to be limited at elevated temperatures unless high levels of inhibitors are used. Such high levels of inhibitors, however, will increase cure times and can even result in incomplete cures. Ashland has done extensive studies of type and level of inhibitor as stability during the past five years. This has led to the development of several unsaturated polyester-styrene resin systems stable in excess of two years when stored at 70-80 F.

The unsaturated polyester resins exhibited high physical strength properties in the screening tests. Since the resins will be subjected to elevated temperatures in the field, stability tests were run at elevated temperatures using off-the-shelf type inhibitors. The effect of resin type, styrene level, and inhibitor type and level were checked. Test temperatures of 158, 130, and 110 F were used. The data is listed in table 5. The resins as listed are in the order of decreasing activity. That is, Aropol 7010, listed first, is the fastest curing and Aropol 7720M is the slowest.

Generally, the fastest reacting unsaturated polyester resin system had the least stability. Increasing the styrene level from 60% to 70% also reduced the stability noticeably. This is apparent with most of the resins tested. For example, using 0.25% dimethyllauryl-amine-hydrochloride (IMLA-HCl) as the inhibitor, Aropol 7010 at 40% resin solids had 55-day stability at 130 F. Reducing the solids to 30%, the resin remained fluid for only 19 days at 130 F. The same trend was noted with Aropol 7510M. Using 0.5% of Adogen 464, a quaternary amonium chloride, as the inhibitor, Aropol 7510M remained fluid 150 days at 130 F. When resin solids were reduced to 30%, the resin system remained fluid only 79 days at 130 F.

Temperature has a very pronounced effect on the stability: the polyester stability is 25 days at 158 F; whereas at 130F, 150-day stability was obtained with the same inhibitor. Some inhibitors provide less stability if too high levels are used, e.g., Aropol 7510M at 40% resin solids was evaluated with dimenthyllauryl amine hydrochloride as the inhibitor at 1, 0.75, and 0.5% levels. At 158 F, no difference was noted in stability. However, at 130 F, the resin with 1% DMIA-HCl gelled after 101 days; with 0.5% inhibitor, it gelled after 111 days; and with 0.25%, it was fluid for 130 days.

The best inhibitor in the test series was a blend of Adogen 464 with copper naphthenate. Aropol 7510M at 30% resin solids in styrene had 30-day stability at 158 F, and 160-day stability at 130 F using 0.4% Adogen 464 and 0.01% copper

naphthenate. The next best inhibitors were Adogen 464 alone and dimethyllaurylamine hydrochloride which gave 150 and 130-days stability, respectively, at 130 F at 0.5% use level.

Many of the resin systems tested had not gelled at the termination of the test, but a solid formed within the container, generally on the cover above the resin layer. This was undoubtedly polystyrene. Previous testing by Dow Chemical has shown that styrene will vaporize and, consequently, polymerize in the vapor state in the absence of inhibitors. They have found that by using containers with various liners, such as Heresite, the styrene wets the liner and returns back into the solution without polymerizing.

### Urethane Elastomers

The new QMR, which was issued August 1966, requires that the dust palliatives be effective for a minimum of one month in areas subjected to traffic by ground vehicles or aircraft. With the usage limit of three pounds per square yard or 0.45 gallons per square yard maximum, it was apparent that all previously screened resins would not withstand traffic at these usage levels without rupturing when applied on loose silt, clay or sand.

To bear traffic on loose sand at those usage levels, it is felt that the coating must be elastomeric with the following properties:

- . Elongation upwards of 1000-1500%
- . Tensile strength over 1000 psi

Urethane elastomers are known to possess these qualities. Published data show that urethane elastomers can be formulated with elongations of up to 1300% and tensile strengths more than 8000 psi. However, it is generally noted that the higher elongations are obtained at the expense of the tensile strength. Also, urethane elastomers with these properties are generally cured by vulcanization or post-cured at temperatures in excess of 100 C. This, of course, is impractical for the intended use as dust control agent.

Hased on requirements of the Army QMR, research was redirected from screening and modifying off-the-shelf resins to synthesizing wrethene elastomers that would have the high elongation properties, yet curs at ambient temperatures. Table 6 lists the tensile strength and elongation of some of the elastomers that have been prepared.

Several of the elastomers prepared initially had high elongation but low tensile strength, or high tensile strength but low elongation. One elastomer, a 2000 molecular weight polyether diol-toluene diisocyanate prepolymer at a 3.5/1 isocyanate to hydroxyl (NCO/OH) ratio, had a tensile strength of 2050 psi with an elongation of 870% when cured by atmospheric moisture. Other elastomers of this type, using various catalysts for moisture cure, had tensile strengths ranging from 825 to 1275 psi with elongations ranging from 880 to 950%.

Since curing by atmospheric moisture can result in erratic cure times dependent on atmospheric conditions, similar results were attempted by polyol cure. The polyol cure could be controlled more readily by catalysts. However, the results with the TDI prepolymers were discouraging because the proper balance of tensile strength and elongation could not be achieved.

To obtain higher elongations, emphasis was changed from the aromatic, TDI, to aliphatic diisocyanate-based prepolymers. With the aliphatic diisocyanates, aromatic amines can be used in the cure to obtain more rigidity, if desired. The aliphatic isocyanates are less reactive than the aromatic, and whereas the reaction of an aromatic isocyanate with an aromatic amine is instantaneous, the aromatic amine with an aliphatic isocyanate is slowed down enough so a complete mixing can be accomplished. Aliphatic amines are too reactive even with aliphatic isocyanates to obtain good mixing.

Two aliphatic diisocyanates were used: General Mill's DDI, dimer diisocyanate, and Mobay's HX, hexamethylene diisocyanate (HDI). Using DDI and a 2000 molecular weight polyether diol as the prepolymer, elongations of 800-900% were obtained with various aromatic diamine-diol blends as curing agents. However, the tensile strength was low at 200-300 psi.

Several HDI-2000 molecular weight polyether diol prepolymers were prepared, varying the reaction conditions (reaction time, temperature, additions, etc.). The isocyanate to hydroxyl ratios were also varied from 3/1, 2.5/1 and 2/1. At the higher NCO/OH ratios, the elongations were generally lower than desired (~500%). Very encouraging results have been obtained using an NCO/OH ratio of 2/1. Using varying ratios of aromatic diamine-diol blends for curing, urethane elastomers with over 2200% elongations and tensile strengths

over 1000 psi have been prepared. The maximum elongation and tensile strength could not be determined on many of the elastomers on the Instron Model TT-C Testing Machine because the elongation was beyond the limit of the machine. Some of these were subsequently tested on the Scott Model L-6 Tensile Tester to measure the ultimate elongation.

Using a blend of a 200 m.w. aromatic diamine and a 52 m.w. diol at an amine-OH molar ratio of 0.25/1, an elastomer with 620 psi tensile strength at 1150% elongation was obtained. The ultimate elongation (at break) was 1820%. Other elastomers tested: 810 psi tensile strength at 1290% elongation (ultimate elongation 1740%) when cured with an aromatic diamine-52 m.w. diol blend at an amine-OH molar ratio of 0.5/1; 825 psi tensile at 1250% elongation (ultimate 1880%) cured with amine-134 m.w. diol-amine-OH molar ratio 0.5/1; 690 psi tensile strength at 1230% elongation (ultimate 2240%) cured with amine-52 m.w. diol at amine-diol molar ratio 0.25/1.

Storage stability of isocyanate-terminated prepolymers is achieved by using trace amounts of benzoyl chloride. Benzoyl chloride acts as a non-aqueous buffer that maintains a slight acidic environment. This prevents an excessive degree of cross-linking which is obtained when catalyzed under basic conditions, thus affecting the tensile strength development to a slight degree, but ensuring a high elongation.

The last nine elastomers in table 6 can be used to compare the properties of the elastomer vs the benzoyl chloride level. Series #598-600 prepolymers were prepared with 0.04% benzoyl chloride. The next three (601-603) were prepared from a prepolymer with 0.02%; 0.01% benzoyl chloride was used to prepare series 604-606. Comparing physical strengths of the three prepolymer-based elasuomers cured with the same curing agent (diamine-52 m.w. diol at 0.25/1 molar ratio) (#598, 601 and 604), it can be seen that the elastomer with the highest benzoyl chloride in the prepolymer (#598) had the lowest tensile strength, 630 psi. The difference in the tensile strength between the elastomer from the 0.02% benzoyl chloride prepolymer (#601) and the 0.01% benzoyl chloride prepolymer (#604) was not as noticeable (811 vs 895 psi). The same trend is noticeable when using a blend of diamine-90 m.w. diol (molar ratio 0.5/1) as curing agent (#600, 603, 606) where the tensile strength varies from 765 psi with the highest benzoyl chloride level to 1310 with the intermediate and 935 psi with the prepolymer containing the lowest benzoyl chloride level.

Several of the elastomers with the higher tensile strengths and elongations were applied on sand and loose, dry clay by spray-on application. The elastomers performed satisfactorily on sand when 1-2% Cab-O-Sil was used as a thixotroping agent. On loose, dry clay, however, the clay had to be pre-wet with water to effect a surface coating. After the spraying was complete, the clay surface formed shrinkage cracks which split the elastomer surface before the cure was sufficient to withstand the shrinkage. The problem was partially eliminated by using an aqueous amine solution for the pre-spray. Using this solution, there was no cracking of the surface, but the elastomer coating pulled away from the edges of the wooden form, forming small cracks along the edges.

However, wetting the soil very lightly with a fine water mist, not enough to cover the surface, and then spraying on the urethane resin with 1/3 to 1/2 theoretical level of curing agent, a very satisfactory surface coating was obtained. Occasionally, very small cracks formed on the edges. Those specimens without cracks passed the wind and water erosion tests.

### PART IV: PHYSICAL CHEMISTRY STUDY

Estimation of Surface-Free Energies of Soils and Correlation of Wetting with Compressive Strength

In order to estimate the surface-free energy of a solid, the contact angles of several liquids must first be measured on the solid surface. The surface-free energy of the solid is then determined by graphical means using the cosine values of the contact angles and the surface tensions of the liquids. The work reported here consists of: 1) measurement of contact angles on sand, silt, and clay; 2) estimation of the respective surface-free energies; and 3) correlation of resin spreading with the compressive strengths of resin-soil samples.

### Contact Angle Measurement

From previous work on the critical surface tension (i.e., an estimate of the surface-free energy) of glass, it was known that only liquid metals and aqueous solutions of inorganic salts such as potassium carbonate and calcium chloride would exhibit contact angles greater than zero on the soil samples. The first phase of this work consisted of compressing the soil samples into slugs in a hydraulic press, removing the slug from the press, and measuring the observed contact angles of mercury and aqueous solutions of potassium carbonate as a function of applied pressure. It was anticipated that constant contact angles would be obtained with increasing applied pressure. Good results were obtained with mercury as shown in figure 1. The soil slugs were much too porous, however, for contact angle measurements using aqueous potassium carbonate solutions. Penetration was almost instantaneous and no reliable contact angles could be obtained.

An alternate method is to make use of the Washburn equation which predicts the rate of rise of a liquid into a capillary:<sup>2</sup>

$$\frac{d1}{dt} = \frac{2r \gamma \cos \theta - r^2 \log g}{871} \tag{1}$$

D. A. Olsen and A. J. Osteraas, J. Phys. Chem., 68, 2730 (1964).

<sup>2.</sup> See, for example: J. T. Davies and E. K. Rideal,
"Interfacial Phenomena," 2nd Ed., p. 423, Academic Press,
New York (1963); L. I. Osirow, "Surface Chemistry,"
p. 277, Rheinhold Publishing Corp., New York (1962).

where: 1 = length of penetration in time t

r = capillary radius

7 = surface tension of the liquid

O = contact angle

~ = density of the liquid

g = gravity

r = viscosity of the liquid

For very small capillaries such as those in a bed of packed soil the second term is small and Washburn's equation reduces to:

$$\frac{d1}{dt} = \frac{r \cdot / \cos \theta}{4 \cdot i \cdot 1} \tag{2}$$

which may be integrated to:

$$1^2 = \frac{r \gamma t \cos \theta}{2 \eta} \tag{3}$$

Since the capillary radius, surface tension, contact angle, and viscosity are constant for a given liquid and solid, equation 3 reduces to:

$$1 = k \sqrt{t}$$
 (4)

and where:

$$k = \frac{r^2 \gamma^2 \cos^2 - \Theta}{4 \gamma^2} \tag{5}$$

Thus, the penetration of liquids into the packed bed of the soil samples should be linear with root time. This prediction is shown to be true experimentally in figures 2, 3, and 4 for several liquids on sand, silt, and clay, respectively.

Since the values for the various radii of the capillaries in the packed beds cannot be determined, a tortuosity factor,  $\gamma$ , must be included in equation 3 to give:

$$(\cancel{t})^2 = \frac{\mathbf{r} \, \gamma \, \mathbf{t} \, \cos \, \Theta}{2} \tag{6}$$

or

$$1^2 = \frac{\mathbf{r}}{\sqrt{2}} \cdot \frac{\sqrt{\mathsf{t} \cos \Theta}}{2} \tag{7}$$

For a given packing the bed constant  $r/r^2$  will be constant. The bed constant can be determined by using liquids which wet the solid, i.e.,  $\cos \theta = 1$ . For this case, equation 7 becomes:

$$\mathbf{1}^2 = \frac{\mathbf{r}}{2^2} \quad \frac{\mathbf{t}}{27} \tag{8}$$

For evaluating contact angles from the experimental data of figures 2, 3, and +, the factor  $r/\chi^2$  can be eliminated from equations 7 and 8 to give:

$$\cos \theta_{\hat{1}} = \frac{(\text{slope}_{\hat{1}})^2 \eta_{\hat{1}} \qquad \gamma'_{\hat{2}}}{(\text{slope}_{\hat{2}})^2 \eta_{\hat{2}} \qquad (9)$$

where the subscripts 1 and 2 refer to the aqueous solutions and absolute ethanol, respectively. Absolute ethanol is assumed to completely wet the soil samples (i.e.,  $\cos \theta = 1$ ). The values calculated for the various contact angles are given in table 7. As can be seen from figures 2 and 4, the data for ethanol, water, and 2M calcium chloride have very similar slopes. Since the packing of the columns was undoubtedly non-uniform, any attempt to calculate contact angles from the small differences in these slopes is subject to considerable error. Such data were not used to estimate surface-free energies.

### Estimation of Surface-Free Energies

Once contact angles have been obtained, the normal procedure is to plot cos — versus the liquid surface tensions and extrapolate to zero contact angle and thus obtain the critical surface tension<sup>3</sup> (i.e., an estimate of the surface-free energy). Such a plot is not feasible here because of the widely separated surface tension values of the calcium chloride solutions and mercury. It is possible, however, to use the Gibbs-Autonoff equation, 4 viz:

$$\frac{\cos \Theta}{\sqrt{L}} = \frac{2\sqrt{S}}{\sqrt{L}} - 1 \tag{10}$$

See, for example: W. A. Zisman, Advan. Chem., Series 43, 1, (1964).

<sup>4.</sup> V. R. Gray, Chemistry and Industry, p. 969, June 5, 1965.

which predicts a linear plot of cos  $\theta$  versus  $1/\gamma_L$  with a slope of  $2/\varsigma$  where  $\gamma_S$  is the surface-free energy of the solid. Alternatively, the graph may be extrapolated to cos  $\theta = 1$  where  $\gamma_S = \gamma_L$ . Such a graph for the contact angle data obtained on the soil samples is shown in figure A5. The value obtained for the surface-free energy of sand is 68 dynes/cm, of clay 60 dynes/cm, and of loam 49 dynes/cm. The value for sand is very close to the critical surface tension obtained for glass, viz 72 dynes/cml, and thus is reasonable. Since silt presumably contains the largest amount of organic materials, the lowest value is expected for silt, as was the case, with an intermediate value for clay.

# Correlation of Spreading Coefficients with Compressive Strengths

After the surface-free energies of the various soils have been determined as above, it then is possible to determine whether spreading of a resin on a soil sample will also occur. The spreadings of a liquid on a solid is governed by: 5

$$S = \gamma_{S} - \gamma_{L} - \gamma_{LS}$$
 (11)

where  $\gamma_S$ ,  $\gamma_L$ , and  $\gamma_{LS}'$  are the surface-free energies (i.e., surface tensions) of the solid, liquid, and interface, respectively. For spreading (i.e., wetting): S>0, and for non-spreading: S<0. An approximation can be derived from equation 11 for the case of an organic liquid spreading upon a low energy solid surface, since it is then reasonable to assume that  $\gamma_{SL}$  is negligibly small in comparison with  $\gamma_{LV}$ . Therefore:

$$S = \gamma'_{S} - \gamma'_{LV} \tag{12}$$

and

$$\gamma_{\rm S} > \gamma_{\rm LV}$$
 for spreading. (13)

Hence in all such systems, when spreading occurs, the specific surface-free energy of the liquid is usually less than that of the solid.

<sup>5.</sup> See, for example, A. W. Adamson, "Physical Chemistry of Surfaces," p. 264, Interscience Publishers, New York, (1960).

Scharpe and Schonhorn have stated explicitly that the criteria for spreading and for strong adhesion bear a one-to-one correspondence to each other and that it is, therefore, possible to decide, a priori, whether a liquid will adhere strongly to solid surface if the specific parameters in equation 13 are known. Equation 13 is the fundamental relationship that determines whether strong adhesion will occur.

Thus, for the case of resins on soils, if the surface tension of the resin is known, it should be possible to predict whether the resin will spread on a given soil sample and also, if the resin will adhere.

Several resin systems were chosen and the surface tension of the resin and/or its major components were determined using a duNouy tensionmeter. The results are shown in table 8 along with predictions of spreading and adhesion. The spreading predictions follow from equation 12, where if the value of the spreading coefficient S increases the tendency of the resin to spread on the particular soil (i.e., wetting) also increases. The adhesion predictions were made from the relative magnitude of the spreading coefficients.

After the spreading coefficients were determined, the results were correlated with actual tests of compressive strength from another portion of this study. The compressive strength is presumed to be an indication of the degree of adhesion. Both sets of data are shown below and the correlation is shown in figure 6.

Resin System	Spreading Coefficient dynes/cm		Compressive Strength psi			
Aropol 7110 Styrene	<u>Sand</u> +36	Clay +28	Silt +17	Sand 2070+ 230	Clay 1653+ 114	Silt 1151+ 551
Sodium Silicate N	- 9	-17	-28	<b>32</b> 2	76	30

From figure 6 it is seen that a good correlation was obtained between compressive strength and the spreading

Market Commence of the

<sup>6.</sup> L. H. Scharpe and H. Schonhorn, Advan. Chem. Series, 43, 189 (1964).

coefficients. Two possible explanations for this correlation are: (1) the criterion of spreading will predict adhesion as discussed at length above, and (2) the soil slugs used in the compression tests were made by pouring resin over the soil and allowing it to penetrate into the soil. Thus, the effect observed in figure 6 results from fortuitous differences in the rate of penetration as predicted by equation 8.

More work would be necessary to analyze the correlation of figure 6. If the spreading coefficient does indeed predict the compressive strength, then the effectiveness of new resins could be predicted simply by a determination of its surface tension and a knowledge of the surface-free energy of the soil on which it is to be applied.

### PART V: CONCLUSIONS AND RECCMMENDATIONS

### Conclusions

Based on the resins screened for dust palliatives, soil waterproofing agents and soil stabilizers, the following conclusions can be drawn:

- I. Maximum unconfined compressive strength (>1000 psi) on soils was obtained with epoxy resins and unsaturated polyester resins. However, high use levels (11 and 6 pounds per square yard, respectively) were required to obtain 1 1/2 to 2-inch penetration by pouring the resin on the soil.
- 2. Intermediate levels of soil strengthening (200-1000 psi) were obtained with urethane, alkyd, acrylic, aniline-furfural, sodium silicate and plasticized sulfur resin systems.
- 3. Most of the resins had adequate resistance to water erosion. Notable exceptions were emulsion types such as Coherex, which washed away; the 15XPF gelatin sample, which resisted erosion but became flexible and easily broken presumably due to absorption of water; the aniline-furfural sample, which did not change but absorbed water; and Chem Rez samples, which exhibited some surface erosion.
- 4. Excellent resistance to freeze-thaw and wet-dry cycling was exhibited by composites prepared from epoxies, unsaturated polyesters, polyurethanes, and solvent dilutable alkyds. Samples prepared with sodium silicate, acrylic emulsions, and water soluble alkyds crumbled or spalled during one or both of the cycling tests. Reductions in unconfined compressive strength were observed with AM-9, sodium silicate, aniline-furfural and Emlon E-200. Some of these reductions in strength may not be real, but due to poor sample preparation.
- 5. Maximum water repellency was obtained with moisture-cured urethanes, octadecyl isocyanate, phenol-modified Chem Rez 200, and solvent system alkyds. The unsaturated polyester resins gave "fair" to "good" water repellancy.
- 6. Based on economic considerations as well as performance results, the unsaturated polyesters with high levels of atyrene and moisture-cure urethanes appear the most promising for soil treatment providing sufficient quantities can be used to develop strength. For water repellency without

dust palliation and soil strengthening, octadecyl isocyanate is the most effective. Areas not subjected to traffic in any form can be effectively treated with air-dry alkyds.

- 7. Although many of the other resin systems tested could be effective for soil stabilization and/or dust palliation and soil waterproofing, they were found unsuitable due to slow cure times in excess of four hours or to erratic cures.
- 8. Unsaturated polyester resins exhibited high flexural strengths when mulled with sand. Values of over 2200 psi flexural strength were obtained using 20% of sand weight of a 30% resin solids Aropol 7510M in styrene.
- 9. Maximum elevated temperature storage stability on unsaturated polyester-styrene resin systems was obtained with a blend of 0.4% Adogen 464, a quaternary ammonium chloride, and 0.01% copper naphthenate (percentages based on resin weight).
- 10. Based on observations of field tests and on the requirements in the QMR issued August 1966, it is felt that for the resin to be effective for use in traffic areas at the low use level, the resin must have high elongation over 1000% and high tensile strength over 1000 psi.
- 11. Several urethane elastomers have been prepared with elongations over 2200% and tensile strength over 1000 psi.
- 12. Sand treated with the urethane elastomers passed the 150-mph air impingement and water erosion test. A flexible surface coating was formed.
- 13. When the urethane was applied on loose, dry clay, coverage was poor and shrinkage cracks formed in all the resins tested.
- 14. Wetting the surface with a very low level of water, not enough to cover the surface completely, and then applying the urethane resin with only 1/2 to 1/3 of the theoretical amount of curing agent resulted in a very satisfactory surface. These specimens passed the 150-mph air impingement and water erosion tests.

## Recommendations

Urethane elastomers have shown excellent promise as treatment materials for dust palliation and soil waterproofing. Formulation changes can be made to achieve high elongation and high tensile strength. Research should be conducted to obtain elongation of over 2000% with tensile strength of 2000 psi. Several factors, which affect both elongation and tensile strength, should be evaluated. Among these are:

Type of isocyanate used Type of polyol used Reaction conditions during polymer preparation Curing agent used

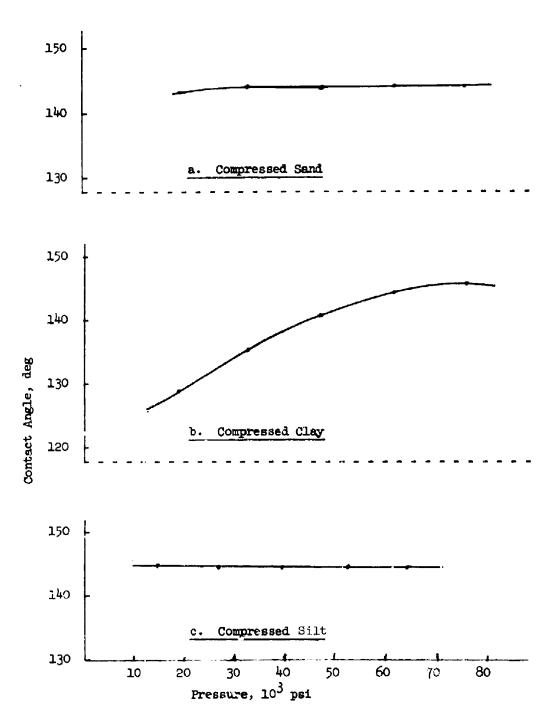


Figure 1. Contact Angles of Mercury

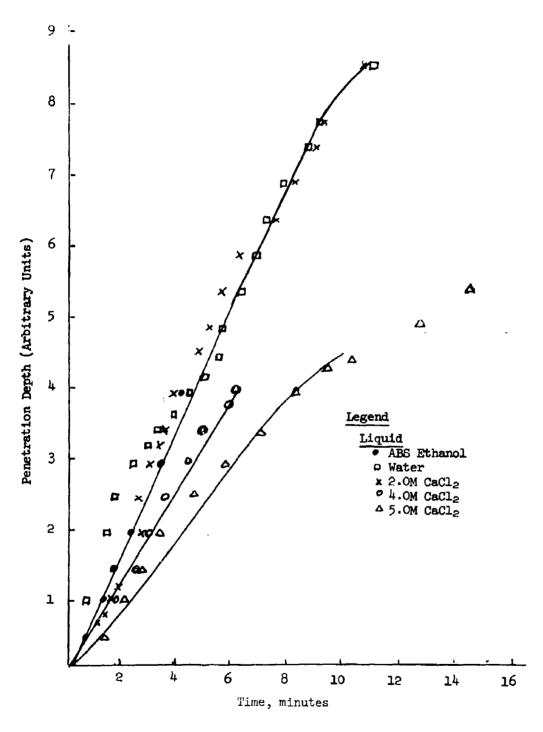


Figure 2. Liquid Penetration into Sand

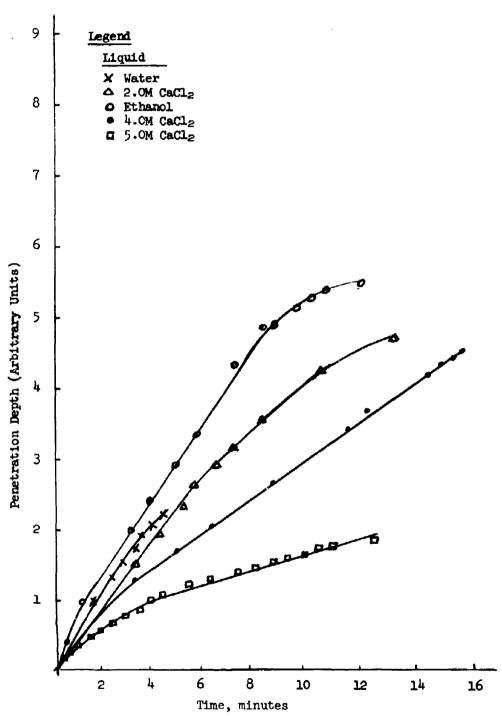


Figure 3. Liquid Penetration into Silt

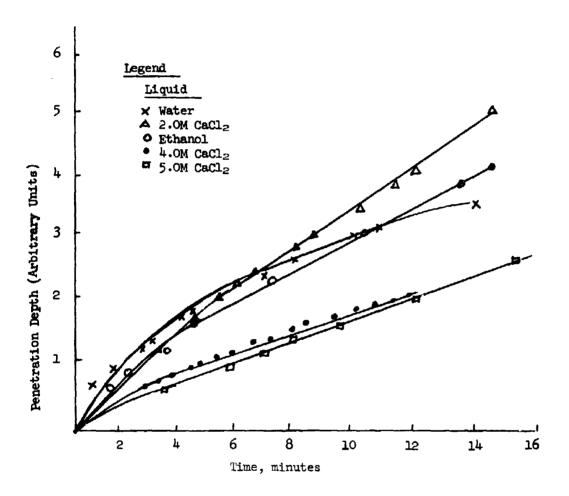


Figure 4. Liquid Penetration into Clay

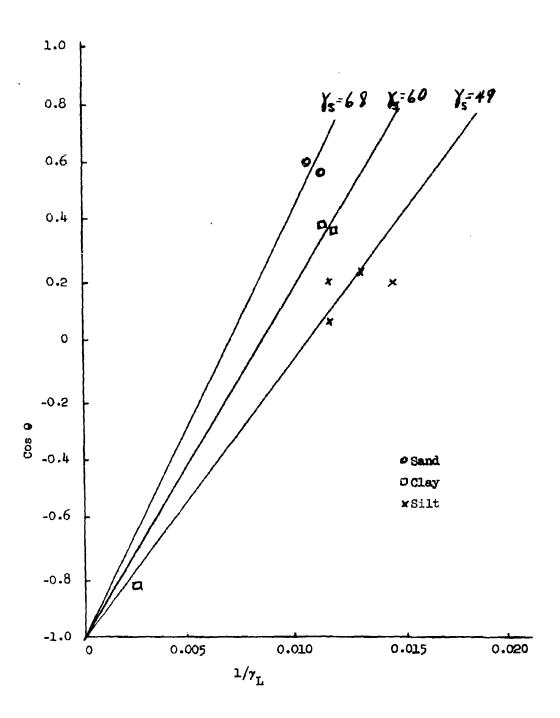


Figure 5. Estimation of Surface-Free Energies

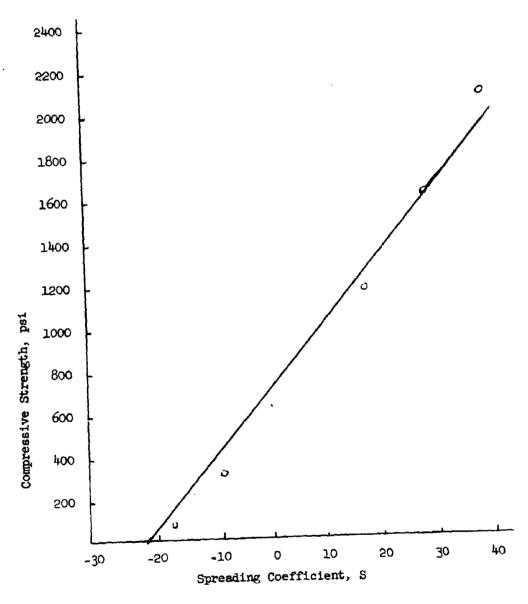


Figure 6. Correlation of Spreading Coefficient and Compressive Strength

							Let	1	st	ist																		ier	ě	_	-
		8 × 1 5 E E	Crust erumbly	Hard cake	Break up easily	Very hand	Did not harden, stayed moist	Did not harden, stayed moist	Did not harden, stayed motst	Did not harden, stayed moist		Hard, flex, easily broken	Flexible-porous	Crumbly, soft	Surface skin	Surface sain	Surface akin	Surface skin	Surface skin	Surface skin	Surface akin	Surface skin	Hard crust, crumbly	Surface skin, no strength	Surface skin	Surface skin	Surface skin	Soft, partly eroded by water	Soft, pertly eroded by water	Stayed wet. did not harden	Stayed wet, did not harden
	Water	Repellency						[4,	[s.					D.	ŗs.	- 64	M	ĵa,	[Bac	ía,	(e)	(A)	۵.	Œ.	-	<b>(2.</b>	Sa,			•	ß.
	Curing Penetration	In.c	0.25	1.5	2.0	5.0	1.5	5.0	0.25	1.5	0.5	1.5	1.5	Thu:	0.5	Ę	TTI	1:25	0.5	5.0	0.25	0.25	T)	T True	2.0	1.0	1.0	1.5	1.5	5.0	5.0
	Curing 1	Time, Mr	3	57	ដ	ដ							о. •	ţ	0.25	0.25	0.25	4	ų,	ţ	0.1	0.1	å	<b>*</b>	å	ţ.	÷č.	₺	<b>.</b>	å	\$
	Water	Content, &	È	È	2	È	È	9	្ន	Ē	Ē	È	ŝ	01	or	٥Ţ	91	2	9	2	2	ន្ទ	Q	30	នុ	នុ	្ន	È	È	10	ន
5011		Compaction	Loose	Loose	Loose	Loose	Loose	Loose	Loose	Loose	Loose	Loose	Loose	Compact	Loose	Loose	Loose	Locae	Topec	Loose	Loose	Loose	Compact	Compact	Loose	Loose	Loose	Loose	Loose	Loose	Loose
		E.	Sem	Send	CLEY	P.	Sand	Serie S	C)	Sand	Send	Serie	Sent	Send	Sernd	i S	311	Sand	ð	S11¢	i i	311	Sen	<b>Send</b>	Send	ğ	31,	Serie	Sand	Serre	S.
olvent	Application	Rate, 1b/yd2	5.1	14.4	35.5	15.4	80 80	9.4	9•	8.8	e:	9.71	8.8					9.4	9.	9.4					6.1	9.1	9.1	5.4	3.6	9.1	9.1
801		Type	xolox	Solox	Solox	Solox	Water	Water	Water	Water	Water	Water	Mater	Water (2.5%)	None	None	None	Water	Water	Water	None	None	None	None	Water	Water	Water	Solox	Solox	Solox	Solox
		Catalyst	Ashland S-1496	Ashland S-1496	Ashlend S-1496	Ashlend S-1496	Mone	None	None	None	$cr_2(30_{\mu})_3$	Formalde de	DWAPH-KSe	DMAPH-ICPe	None	3C1.2	Nore	None	None	H,PQ.	M <sub>3</sub> PO <sub>4</sub>	M <sub>3</sub> No.	M <sub>3</sub> P0.								
••	Application	Rate, 1b/yd2	1.3	5.5	5.5	о. П	6.0	9.4	9.	2.9	1.3	₽. <del>1</del>	2.1		1.6	9.1	9.1	9.1	9.1	9.1	9.1	9.1			9.1	1.6	9.1	2.3	4.6	3.6	3.6
Bestn		Type	Spon 828				Cohere::			Ashlerd Emul	15 XPF Gelatin		6-5	NN-9(3.04)	EP 8908-23					•	EP 8908-122	,	EP 8908-122	(0.09/- EP 8908-122 (8 set 1	906-129			Chem Rez 200		Phenol Mod	
	ã	ģ	4	5	Ä	<b>5</b> 4		6 21		-		គ	ă	8			E	3	<b>%</b>				_	98	-	<del>بر</del>		17A C	<b>164</b>	9	, .ag

a. Application rate of reals includes catalyst.
 b. Water content based on dry weight of soil prior to addition of reals.
 c. Dapth of penetration of reals into soil. Mail refers to blending reass into soil.
 d. Approximately 0.5 ml of water added to surface 2 hr after reass was applied. Effectiveness of reass rated as follows:

E = Excellent, water as droplet after 10 min. G = Good, water visible on surface after 5 to 10 min. F = Firt, water visible on surface after 1 to 5 min. P = Foor, water absorbed in 1 min or lass.

e. Percentage based on dry weight of soil. Catalyst weight included in percent resin.

1 of 4 sheets

		** * * * * * * * * * * * * * * * * * *	Hard crust	Hard crust	Hard crumbly edges	Thin surface shell		Absorbed slowly, cracked when dry	Upper 1/2 inch dried and separ-	ated from rest	Very hard	Very hard	Hard, bottom crumbly	Incomplete cure, foamed	Incomplete cure, foamed	Surface hardened, porrous	Surface Mardened, porous	Very hard		Mard, crumbly adges	Breaks up readily	Very hard	Hard, edges countly	Very hard	Solidified on contact with sand	Vary hard, 15 min to penetrate	Vary hard	Wery crumbly edges, 5 min to	penetrate	Very crimbly	Hard surface shell				P	Hard	Very herd		Very hard	Very crumbly	
	Water	Repellency				<u>Ge</u> .	•									Δ.	o.													D.	E4	6	d t	z. 1	<b>5</b> 24,	ы	U	,	<u>ρ</u> ,	ω	
	Penetration		1.5	5.0	1.25	mu]]		5°.E	0.5		5.0	ر. د.	1.5	mul 1	mL1	0.5	0.25	TTTT		7	겉	TTP1	Ţ	T T	TTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTTT	1.5	o.5	1.5		T(Trum.	mul 1		2			1.5	6.		o N	5.0	
	Curing	Time, Er	÷	ų	ţ,	å		1.5	0.0	ı	7.0	1.5	1.5			0.4	0.1	o.		0:	0.1	0.1	<b>7.</b> 0	0.1		ţ,	5.0	ţ,		ŧ.	ţ.	ć	Ļ			0.4	ψ		t,	<del>,</del>	
	Water	Content, \$	È	ğ	È	ç	ì	È	À	•	È	È	Ē	Ē	È	9	2	Ē		Ē	Š	ğ	Ē	È	Ē	È	È	È		01	O‡	5	3 5	2 :	2	2	ន្ទ	;	oj	07	
8011		Compaction	Loose	Toose	Poor	Comment		10086	Locae		Loose	100	Pose	2000	Loose	Loose	Loose	Loose		10086	Loose	Toole	Loose	Loose	Tool e	Loose	Loose	Loose		Compact	Compact		2	10086	Loose	Loose	Loone		Loone	Loose	
		Ž,	Cley	Cley	Sit	Į.	<u> </u>	Gund	C)		71 87	C)	311	Send	Sent	Sand	1	Send		<b>3</b>	311	98 nd	9	캶	11 8	Send	S	Silt		Send	Sand			3	311	3	Send	1	200	311	
olvent	Application	Rate, 1b/yd	1.6	4.3	S			10.t	19.0	: }	16.4	9	53.6	•		9.4	4		(33)	(S)	(6.74)					15.3	11.5	8		1000	( to	(6)	÷.		m.	3.6	 	4	7.3	7.3	
SOLV		type	Solox	Solos	Solox	e lo	(1.36)	Water	Unter		Water	Vater	Vater	None	None	Water	Water	Water		Water	Water	None	None	None	None	Styrene	Styrene	Styrene		Styrene	Styrene	ě	Scyrene	Styrene	Styrene	Styrene	TAT.	Acetate	Ethyl Acetate	St.ny.1	Acetate
		Catalyst	37 <b>5</b> HC1		324			DETA	É	i	DETA	DETA	12 <u>8</u> 0	Maconate 4040	Meconete 4040	MasS17	Ma-31F	Maz 31 F.	•	MazS1F	Na <sub>2</sub> S1F	E-13	E-13	E-13	<b>E-1</b> 3	BEO-DIM	BPO-DM	BPO-DW		RPO-DIM	BPO-DMA	1	10.01.01.01.01.01.01.01.01.01.01.01.01.0	HO-DA	320-DW	320-DM	None		None	None	
<b>€</b> c	Application	Rate, 1b/yd <sup>2</sup>	3.0		) -1	-		6.9	7.0	<u>:</u>	10	19	10				감									3.6	3.7	6.6							7.3	7.3	7.3		7.3	3.3	
4 C		Type	Analine-			4 14 14	Purfural (6.741)	Carlon E200						P-2010(22.34)1	P-2010(33,5%)1	Na Silicate N		Ne Stlicate N	(214)		•	Sul nur (164)		•	Sulfur(23.65)1	Aropol 7110				Aropol 7410	tropol 7110		možet (Kom				Arothane 160				
	2	ż		4.0	: ::				2	į	7	Q.	~		_	7				~								Ą		ዴ	257	9	3 :	₹.	3		<u>.</u>		<b>∄</b>	<b>1</b>	

2 of 4 sheets

Table 1 (continued)

		AALSEAK	Mart but crumbly	Bard but cruebly		Surface penatration only, and	CHAINCE PERSONALISM ONLY, MALE			Section of the sectio	State of the	Tibre but			Party Bull acc	abrasion fair to good	Mard surface, crushly beneath		Mard Burface, slow cure, crumbly	Crustoly Distributes about	Account many	Surface bardens	Poor cure	Por cure	Ward surface, wet beneath	Mard surface, wet beneath	Crumbly	ery marc	Very bard		Mard Curface, uncured beneath Bathers confere account beneath	M. A. see headen		Did not harden		, , , ,	Arry hard	Slow cure	Nard	
	Mater	Repellency	E I			ம		<b>2. B</b>	<b>.</b> 6		•		-		4 4	4	愛		<b>д</b>	•						_	<b>₽</b> 1		20		F) F	4 6	<b>a</b>	3	9	<b>.</b>	>	s m	ω	
	Penetration		5.0	0.1	:	0.2	· ·	•		,	() ()	CT:0	•	7.0	(2:5)		CLUT		Tank J	0.1	۲۰,	-	5.75	0.25	6.0	1.25	1.5	~	0.5		٠. د. د.		u	21.0	c	J	6.5	Q.	1.25	
	Curing	Time, Fr	ŧ.	đ	;	t.	t i	t d	ŧ,	t d	n d	t d	<b>.</b>	ů á	₹.	ŧ	*		*		3	• -	4 &	t	ŧ	<b>*</b>	0.5	ţ.	ŧ		đ.		<b>C</b> : <b>7</b>	ŧ.	ĕ	t	٥.	ο,		
	Water		10	Ů.	2	ន្ត	0, 5	2.5	9 9	3 :	3;	<b>4</b> ;	3 '	v.	^;	9	97		9	9:	3,5		3 5	ន	2	2	ន	20	ot		99	3,5	3	9	٤	à	ឧ	70	01	
S 0 1 1		Compaction	Loose-	Compact	Compact	Compact	Compact	1007	Compact	Compact	Compact	Compact	100	Loose	Loose	Compact	Compact	•	Compact	Toose	1003	100ge	Toose	Toone	Loose	Loose	Loose	Loose-	Compact Looms-	Compact	Troise	Toole	anoor	Loose-	Compact	Dominact.	Loose	Loose	Lose-	
		Š	Send	į	2	311	9	- C	2	e de	S C	Send of	200	2	SITE	S S	SLIt		Cles	Series (	٠ ا		1 8	2	Series	ð	DE SE	Send	200	}	Send	1	Bung	C1.	į	3	Clay	Sand	Ü	
Solvent	Application	Rate, 10/yd	6.8	ú	?	3.5	3.5	w.	ı, ı	o ·	in.	on i	2.5		2.5	( × )	(dr)	( <del>*</del>	( ph)	9.	9,1	, (v (			0.5	9.4	2.3	3.7	2.7	5	<b>-</b> 1	-, ·	ن •	9.	•	'n	3.7	1.7	1.1	
301v		Type	Ethy]	Acetate	Acetate	Acetone	Acetone	Acetone	Acetone	Acetone	Acetone	Acetone	Acetone	Acetone	Acetone	Ethyl	Ethy1	Acetade	Styrene	Water	Water	Vater	Marer		Water	Water	Water	1243	Acetate	Acetate	Gasoline	Kerosene	Stry1	Fthyl	Acetate	Stroi.	Ethy	Ethyl Acetate	Ethy1	Acentine
		Catalyst	DETTO		AUTHU	1980 1980	AGBA:	None	None	None	<b>200</b>	TABOA	None	TMBDA	TABLE	None	None		None	None	None	None	None	No.	None	None	None	None	ou ch		None	None	Kone	None		None	None	<b>7</b> 2.	<b>72</b>	
	1013	Rate, 1b/yd <sup>2</sup>	3.1		3.1	1.1	1.1	<b>د.</b> ۶	5.0	0.5	0.6	٥٠٥	3.0	3.0	3.0					9.1	9.1	·	۰.	d -	1 0	9.4	7.6	3.7			Ł.5	4.5	4.6	9.4		3.7	3.7	5.0	5.0	
•_	Applic	Rate																																						
X C S 1 D	o) I day	Tope Rate	ه ا			Arothene 170										Arothene 1/0	0/ (4/4)	(35)		Arlon 110		Arlon 310	,	Arion 303	083 0014	30 1511		Aroplaz 832					Aroplez 6003			Arople: 6065		Arothene 170-C. NCO	80	

Table 1 (continued

Street	Res	n 4		5011	Solvent		5011					
stee 1b/yd²         Type Same Content, f° Time, H° T		Application			.pplication			Mater	Curtre	Penetration		
4.5         Sand         Loose         10         24         2         P           9.1         Sand         Loose         10         24         2         P           7.1         Sand         Loose         10         2         P         P           7.1         Sand         Loose         10         2         P         P           6.3         Sand         Loose         10         2         P         P           5.0         Clay         Loose         10         2         P         P           2.0         Sand         Loose         10         2         0.5         E           1.8         Sand         Loose         10         2         0.5         E           1.3         Clay         Loose         10         2         0.5         E           1.3         Clay         Loose         10         2         0.5         E           2.1x         Loose         10         2         0.5         P           3and         Loose         10         2         0.5         P           Clay         Loose         1         2         P         P <th>3 X 2 C</th> <th>Rate, 1t, yd</th> <th>Catalysi</th> <th>Type</th> <th>Rate. 15/yd2</th> <th>7</th> <th>Compaction</th> <th>Content, \$</th> <th>7 m</th> <th>In.c</th> <th></th> <th></th>	3 X 2 C	Rate, 1t, yd	Catalysi	Type	Rate. 15/yd2	7	Compaction	Content, \$	7 m	In.c		
7.1         Sand         Loose         10         2*         2         6           7.1         Sand         Loose         10         2*         2         P           7.1         Sand         Loose         10         2*         2         P           7.1         Sand         Loose         10         2*         2         P           6.3         Sand         Loose         10         2*         2         P           2.0         Olayer         10         2*         2         P           2.0         Sand         Loose         10         2*         0.5         E           1.8         Sand         Loose         10         2*         0.75         G           1.3         Clay         Loose         10         2*         0.75         E           2.0         Sand         Loose         10         2*         0.75         P           1.3         Loose         10         2*         1,         P           Clay         Loose         10         2*         0.75         P           Clay         Loose         10         2*         1         P     <	flint 505	3.0	None	Styrenc		70 S	Loose	억	5		۵,	No cure
1.1         Sand         Lose         10         2+         2         P           7.1         Sand         Lose         10         2         2         P           7.1         Sand         Lose         10         2+         2         P           6.3         Sand         Lose         10         2+         2         P           5.0         Clay         Lose         10         2+         2         P           5.0         Sand         Lose         10         2+         0.5         E           1.8         Sand         Lose         10         2+         0.75         E           1.3         Lose         10         2+         0.75         E         C           1.3         Lose         10         2+         0.75         E         C           1.3         Lose         10         2+         0.75         E         C           1.3         Lose         10         2+         0.75         P         C           1.4         Lose         10         2+         0.75         P         C           1.4         Lose         1         2+	a.mo-o	۲۰۶	None	Hineral Seining		Sand	Loos	07	<b>*</b>	Ci.	ی	Stayed wet, no cure
1.1         Sand         Loose         10         2         2         P           1.1         Sand         Ioose         10         2*         2         E           6.3         Sand         Ioose         10         2*         2         P           5.0         Clay         Conyect         10         2*         1         E           2.0         Sand         Loose         10         2*         0.5         E           1.8         Sand         Loose         10         2*         0.75         C           1.3         Clay         Loose         10         2*         0.75         E           1.3         Loose         10         2*         0.75         E           Clay         Loose         10         2*         0.75         E           Clay         Loose         10         2*         0.75         P           Clay	A 970 0	2°C	ilone	Mineral		Pi es	Icose	10	Ć.	۲,	P4	Stayed wet, no cure
7.1         Sand         Iocee         10         2         2         E           6.3         Sand         Locae         10         2*         2         P           6.3         Sand         Locae         10         2*         1         E           2.0         Chay         Locae         10         2*         1         E           2.0         Sand         Locae         10         2*         0.75         E           1.3         Chay         Locae         10         2*         0.75         E           1.3         Chay         Locae         10         2*         0.75         E           1.3         Chay         Locae         10         2*         0.75         F           1.4         Chay         Locae         10         2*         0.75         F           1.4         Chay         Chay         Locae         10         2*         0.75         F           1.5         Chay         Chay         Chay         10         2*         0.75         F           1.4         Chay         Chay         Chay         10         2*         0.75         F	3 ama-o	o. <b>2</b>	Nore	Spirits Ethil		pu <b>s</b> S	Loose	10	~	~	Δ,	Stayed met, no cure
6.3         Sand         Lose-         10         2+         2         P           6.3         Sand         Loose-         10         2+         2         E           2.C         Object         10         2+         1         E           2.C         Sand         Loose-         10         2+         0.5         E           1.3         Clay         Loose-         10         2+         0.75         C           1.3         Clay         Loose-         10         2+         0.75         P           Sand         Loose-         10         2+         0.75         P           Clay         Loose-         10         2+         0         P		2.0	Ethylene Glycol	Acetate Ethyl		Sand	Loose	10	۲۰	م	ы	Surface hard, rest loose
6.3 Sand Loose- 10 2+ 2 E  2.0 Clay Compact Compact Compact 2.0 Sand Loose- 10 2+ 2 E  1.8 Sand Loose- 10 2+ 0.5 E  1.9 Clay Loose- 10 2+ 0.75 C  2.0 Sand Loose- 10 2+ 0.75 C  2.1 Sand Loose- 10 2+ 0.75 P  Clay Loose- 10 2+ 0.75 P	no-curre alth	4	None	Ethyl		Sand	Locse	ន	÷	۵	Δ,	No cure
2.c Sand Loose 10 2+ 1 E 2.c Sand Loose 10 2+ 0.5 E 1.8 Sand Loose 10 2+ 0.75 C 1.0 Clay Loose 10 2+ 0.75 C Clay Loose 10 2+ 1, P Clay Loose 10 2+ P Clay Loose 10 2+ P	ا <b>برد</b> ه		Deter	Acetate Ethyl		Sand	Loose-	10	÷	a,	ы	Did not harden
2.0 Sand Loose 10 2+ 0.5 E  1.8 Sand Loose 10 2+ 0.75 G  1.3 Clay Loose 10 2+ 1. P  Sand Loose 10 2+ 1 P  Sand Loose 10 2+ P		5.0	DBCDA	Ethy)		Clay	Loose-	10	*	7	ш	Did not natder
2.7 None Rthyl 1.8 Sant Loose 10 2+ 0.75 G  3.7 None Rthyl 1.3 Clay Loose 10 2+ 0.75 G  3.7 None Rthyl 1.3 Clay Loose 10 2+ 0.5 E  3.7 None Rthyl Clay Loose 10 2+ 1, P  3.7 None Rthyl Clay Loose 10 2+ 0.75 P  2.2 Siroc 2-3 Water Clay Loose 10 2+ 0.75 P  2.3 Siroc 2-3 Water Sand Loose 10 2+ P  3.4 Acetate Clay Loose 10 2+ P  3.5 Siroc 2-3 Water Sand Loose 10 2+ P	23/ 8	3.5	BPC-DKS	Acetate Styrene		Sernd	Compact Loose	10	÷	5.0	ы	Rubbery skin
3.7         None         Extra         1.3         Clay         Loose         10         2         0.5         E           3.7         None         Exty1         Sarid         Loose         10         2+         1,         P           3.7         None         Exty1         Clay         Loose         10         2+         1,         P           2.2         Suroc 2-3         Water         Clay         Loose         10         2+         1         P           2.2         Siroc 2-3         Water         Sand         Loose         10         2+         1         P	1, 101 -101	2.	None	Ethy.	8.4	Sent	Loose	10	\$	0.75	o	
3.i         None         Ethyl         Sand         Louse-         10         2+         1.         P           3.7         None         Arctate         Clay         Loose-         10         2+         0.75         P           2.2         Siroc 2-3         Water         Clay         Loose-         10         2+         1         P           2.2         Siroc 2-3         Water         Sand         Loose-         10         2+         2         P		:,	None.	Acctate Ethyl	1.8	Cl.ev	Loose	10	N	6.5	eu)	
3.7 None Ethyl Clay Loose- 10 2+ 0.75 P  2.0 Suroc 2-3 Water Clay Loose- 10 2+ 1 P  Compact Compact 10 2+ 1 P  Compact Compact 2-2 Suroc 2-3 Water Sand Loose- 10 2+ 2 P		ÿ.	None	Acetate Ethyl		Sand	Loose-	10	<b>5</b>	á`	Δι	Very hard
2.2 Siroc 2-3 Water Clay Loose- 10 2+ 1 P Compact Compact 2.2 Siroc 2-3 Water Sand Loose- 10 2+ 2 P		3.7	None	Aretate Ethyl		<b>₹</b>	Compact Loose-	10	ŧ.	67.0	<u>a</u> .	Very hard
2.2 Stroc 2.3 Water Sand Loose- 10 2+ 2 P	 	C) NJ	Siroc 2-3	Water		CLRV	Loose-	10	<b>5</b>	ч	24	Soft, no cure
		5.5	Stroc 2-3	Vater		Sand	Compact Loose-	10	\$	C)	d.	Soft, no cure

Physical Test Date. Table 2

Unconfined Compressive

Resin   Sein   Solve   Resin   Solve   Resin   Solve   Resin   Solve   Resin   Solve   Resin   Solve				Comple selent	ا نو			יו ביו היווי	
T y P e   Nation   Inches	Solvent				Water E	rosion	l l	Freeze	
Type Rate, 15/24 2  Epon 928 1.3  Sold Sold Sold Sold Sold Sold Sold Sold	14	3011		After		出 8		They 8	_
Epon 828   1.3   2.6   5.2   5.2   5.2   5.2   5.2   5.2   5.2   5.3   1.3	Æ	ě.	Initial	Wind Test	Initial	È	Initial	Cycles	8 c. cles
25.0 25.0 25.0 25.0 25.0 25.0 25.0 25.0	l	No.	351	74.	36	7TE		-	
See		50	1,50	Ç.₹	187	433		•	
Coherex 11.5  Ashland Exp Emul 2.7  15 XPF Celatin 1.3  AN 9 2.1  Chen Rez 200 2.3  Analine-Purfural 3.0  Analine-Purfural 3.0  Analine-Purfural 3.0  Analine-Purfural 3.0		5	1608	1600	1675	1612	1070	83	
Coherex 11.5  Ashland Exp Soul 2.7  13.5 XPF Celatin 11.5  M 9 2.1  Chem Rez 200 2.1  Amaline-Purfural 3.0  M. 9  Amaline-Purfural 1.7  M. 9  M. 9  M. 6  M. 7  M.		Send	1762	1743	1765	1747		•	
11   12   13   14   15   15   15   15   15   15   15		Logn	1622	1614	1645	1505		-	
Coherex 2  Ashlend Exp Smul 2  15 XPF Gelatin 13  AN 9 2  Chem Rez 200 2  Analine-Purfural 30  Analine-Purfural 6  but an E-200 6  1  1  1  1  2  2  4  2  2  2  3  4  2  2  3  4  4  4  5  6		Dard.					c <del>y</del> Cic	•	
11.5 Ashland Exp Swul 5.0 15 XPF Celatin 1.3 AM 9 2.1 Chen Rez 200 2.1 Amaline-Purfural 3.0 Amaline-Purfural 3.0 4.7 4.7 4.7 4.7 4.7 4.7 4.7 4.7 4.7 4.7		Ser.	1653	1623	٤,				
Ashland Evp Srul 2.7  15 XPF Gelatin 11.5  MA 9  Chem Rez 200 2.3  Amaline-Purfural 3.0  4.7  Balon E-200 6.9		<b>72 8</b> 5	1755	1746		v			
5.8  11.5  11.5  11.5  11.5  11.5  11.5  11.5  11.6  11.6  11.7  1		Sard	1613	1671	£.	,			
11.5 15 XPF Gelatin 11.3 AN 9 2-1 Chen Rez 200 4.5 Analine-Purfural 1.7 Balon E-200 6.9		200	1691	1675		1618			
15 XPF Gelatin 1.3  AM 9  Chem Rez 200 2.3  Chem Rez 200 4.6  Analine-Purfural 3.0  4.7  Balon E-200 6.9		DE WS	1713	1705		ů.			
AN 9 2-1 Chem Rez 200 2-3 Chem Rez 200 2-3 Chem Rez 200 2-3 Chem Chem Chem Chem Chem Chem Chem Chem		70 mg	<b>1</b> 33	339	757				
Amaline-Purfural 1,-7  Balon E-200 6,-9  L. 6  L. 6  L. 6  L. 6  L. 7  L		Send	1706	;		1563			
Chem Rez 200 2-3 Amaline-Purfural 3.0 h.7 Balon E-200 6.9			3927	1155	1138	10.17	1723	735	1180
Analine-Purfural 3.0 5.7 5.7 5.7 5.7 6.9 6.9		Send	1031	1028	*	88			
Amaitne-Purfural 3.0 1.7 1.7 1.7 1.7 1.7 1.7 1.7 1.7 1.7 1.7		Sard	1073	101	, 00 00 00 00	B78			
F.7 Balon E-200 6.9		Same	5707	1047	1181	1701	\$	3	ደ
6.9 6.9 6.9		Sand					<u>S</u> ;	105	
Balon E-200 6.9		ខិ					SH:		
6.9		C) Set	1750	1738	1484	06. 1.1	1388	đ	۲.
1. C			1. S	1360	272	121			
MOLTE STREET,	Ni L	Serve	1830	1823	823 1834 1728	1728	<b>65</b> 0	ĸ	540
Magatre 7110 2.6		Serve	1768	3762	1784	1745	+173	1168	1412
0.5		9	82.	1795	1814	1704	23.00	2500	196
1 1/2		9				•	1710	2500	1080
000		Servo					1890	2400	2020
		Sand					<b>18</b>	738	æ Æ

e. Withitood freeze-thaw and wet-dry sycling.
b. Not subjected to water erosion
c. Washed away in 10 winnes
d. Some surface erosion
e. Not subjected to water erosion, weight after air dry
f. Brois after third cycle
g. Crumisla when immersed in water
h. Surface spalled when immersed in water
h. Per cent based on dry weight of soil

									Uncor	nfined Course	2A T 000
e 0:		Solv				Sample Seigi	sample teight, ga			Strength	
	Armed Cartion		Ann lost for				Water E	rotton		Treeste.	
			-	Soll		After		14-84		Het B	¥et.Dy
0 0 X	Pate, 1b/ya	14 Y	Bate, 1b/yd	ž,	Initial	Wind Test	Initial	À	Initial	Cycles	8 cycles
CB3 es [es	- ×	100		]	ł				1	ļ	
200	5	MB COL		2					8	8	*
Arlon 210	· 第一的	None		Send					120		161
Arlon 110	×	au cy		2					į,		;
									9		٠
Arlon 363	, in the second	None		Send					5	166	*
Aronlaz 6008	, X	4-47	¥.0	Parag.					101	36.2	2
2007		: 1		1					•	Í	
Aropast (CO)		JP-4	2.0						8	120,	Si
Aroplaz 832	3.34	4- <b>4</b> -	7.0	Sing					255	2	<b>9</b>
ES- 9006-53	1 N. L	None		3	701	901	100	y¶0	3	(	3
P P308 122	7	1		1	Ì			?	}	•	ξ,
		200									,
2000	×	a dig		200					01.0	15.2	•
1				ļ					277	77	•

Table 3

Spray-On Application of Resins for Dust Pulliation and Soil Materproofing

	Resin	- u							A 2 2 7 7	
6			Application	201		Appearance After	Wind	Water Test.	After	
÷	Type	ļ	Rate, 10,yd	<b>3</b>	Prevet	Cure	Test	gn inc. wt.	Water	Re marks
. <u>1</u>	Cat. Asphalt Fmul.	, V	3.2	Send	2	Good surface	۵.	4531	<b>c.</b>	
.c		÷;	5.6	Sand	5	Good surface	Д,	4224	ĵ.	
g:		29.5	5.3	Sand	9	Good surface	ρ.	-₹3 <u>₹</u>	μ	
4		ň	5.6	Sand	2	Good surface	4	+524	p.	
ai A		17.5	· ·	Jan.	O E	Large shrink cracks	í <b>s.</b>			
E E		36	<b>v</b> .	Clay.	oc.	Shrink cracks	(a.			
<b>6</b> 0		 	3.1	Clay	DIG.	Shrink cracks	ſų			
88		1.5	 	Clay	5	Shrink cracks	,44			
18	5pon 315	ć,	භ	C) a,	no	Many cracks	۵.,			
K 2	50 Hallon 127 00	UT	7.7	Clay	OL	Shrink cracks	μ.			
<b>1</b>	EP 8 408-122	. <del>;</del> }	3.0	Clay	2	Shrink cracks	£e.,			
86	EP 8 003-23	5	0.0	CL M	2	Shrink cracks	ρ.			
20	Arothane 170	ŝ	5	Clay	2	Poor wet and cure	Œ			
98		25	3.6	C S	2	Poor wet and cure	Œ			
EQ	Arothane 160	8	3.0	C ay	yes	Slight foaming	<b>4</b>	4142	۵,	
T.		3	ુ.	Clay	yes	Good wetting, alight	۵,	+238	۵.	
						foam				
<u>e/</u>		જ	3.0	Clay	yes	Slight founding	Д,	<i>₹</i> 41+5	a,	DABCO-Co.
58	Aroplez 6008	33	æ.	Clay	ye B	Good weiting	(te.			
26B		æ	5.4	Cley	8	Poor wetting,				
	,					G18ceraea				
27B	Aroplaz 6065	<del>2</del> 2	2.5	C) ay	yes	Good wetting	<b>ը</b> ,			
g,	Aroplaz 832	<b>5</b> 2	3.0	Cley	8	Poor wetting,	a,	).T +	ſ.	Surface skin buckled in
						physically spread				eind
80 °		33	7. 2.	C18¢	89 20 20	Good wetting	۵.	€5.	α.	Surface skin buckled in wind
8		35	3.0	Sand	2	Soaked in	۵,	+12.	C-	
<b>B</b> **		32	÷.	Clay	yes	Good wetting	۵,	+56>	P.F	Surface skin buckled,
		;	c i	,			,	,	,	sp) ;
358	Aropiez d32-	£	o,	<b>1</b>	Q	Good wetting, soaked in	<b>.</b>	<b>1</b>	<b>0.</b> ,	
ţ.	one serious	35	3.0	Clay	yes	Good wetting	ſæ,			
a C	Aropol 75108	9	6.0	Sand	2	Soaked in	ц	+236	Δ,	
; p:		2	3:1	2 13	92	Small cracks	ω			
Į į		OH.	0.	Clay	2	Resin rich surface	Α,	+563	ւ	
5ca		55	3.0	Sand	2	Soaked in, smell	ы	+210	ρı	
				1		cracks in edges	,			
E.B		20	. લ	C1 ay	yes	Crack on edge	24			Soli blew out of eage out read withheld.

a. Sand used was bluff sand; clay used was dried, pulverized, Mississippi buckshot clay.
 b. Prevet with 1/2 lb/yd water.
 c. Subjected to 150 mpn wind one minute. P = passed, F = failed
 d. Subjected to forced water spray one hour.

2 of 2 sheets

- 1

Table 4

Flexural Strength Data

Aropol 7510 M (30% solid in styrene, 6 wt % resin, 14 wt % styrene) and Dry Sand

Co/INA/H <sub>2</sub> O <sub>2</sub> Co/INA/H <sub>2</sub> O <sub>2</sub> .5/.5/1.5	Water b Repellency	Flexural Strength  3-day Cure 5-day	rength, psi 5-day Cure	Remarks
	EA 4	1087	1654	
	э Э	500 000 000 000 000 000 000 000 000 000	1248	
	1 EQ	1510	1923	
	ы	1380	1902	
	ഥ	3621 1296	1780	
	E	28 1- 1- 1- 1- 1- 1- 1- 1- 1- 1- 1- 1- 1-	1872	
	ធ	9691	2145	
	ഥ	1513	1702	
	田	1534	1769	
	H	1413	1736	
	ា	1680	2017	
	P	1351	1595	
	띰	1700	2163	
	១	1302	1831	
	ធ	1482	1782	
	闰	1361	1920	
	얼	1208	1780	

Based on weight of resin + styrene 9

<u>.</u>

Co = 6% cobalt naphthanate, DWA - dimethyl analine, H<sub>2</sub>U<sub>2</sub> - 50% hydrogen peroxide.
Approx 0.5 ml water added to surface 2 hours after resin application. Effectiveness rated as follows:
E = excellent, water as droplet > 10 min; G = good, water visible 5-10 min; F = fair, water visible
1-5 min; and P = poor, water soaks in < 1 min.

l of 2 sheets

Table 4 (concluded)

	0.403.04.04	10-4-			
Run No.	Co/DWA/H <sub>2</sub> O2	Water b Repellency	Flexural Strength,	ength, psi	Domestic a
	3-3-//	2	7	ama Ama	Venture P
128 <b>3</b>	1.5/1/2	ы	3,1468	าหิลา	
and c	ר/ אַ ר/ אַ ר	ı P	100	4 (	
	T/C.T/C.T	<del>-</del> 1	1184	1553	
130B	1.5/1.5/1.5	E	1768	2047	
1313	1.5/1.5/2	ы	1351	1663	
1323	1/1/1	,	100		
	1/1/1		นี่	SON N	TOW Water in sand
45.7 10.7 10.7 10.7 10.7 10.7 10.7 10.7 10	1/2/2		263	301	
134B	1/3/3		200	0,70	100
1 26 0	0/5/0		- 1	3	1000
1575	2/1/2		007	374	water in
LSOB	2/2/3		315	312	water in
13/B	2/3/1		378	7T†	water in
130B	3/1/3		85 8	215	10% water in sand
139B	3/2/2		2%	ಕ್ಟ	water in
T40B	. 3/3/1		354	343	water in
				•	

Table 5

Storage Stability of Unsaturated Polyester Resin at Elevated Temperatures

7   7   7   7   7   7   7   7   7   7		Insaturated Polyester	Polvester	Inhibitor		Stability	- Days t	o Gel at
Aropol 7010         75         Ashland Proprietary Inhib         0.01         12         6           4.0         Copper majori hante         0.5         11         27         12         13           4.0         Partic majori hante         0.5         20         27         4,6         13           4.0         IndBill         0.75         25         26         27         26         27         27         4,6         13         27         27         4,6         13         27	Pef	Type	Solids, \$	101	Per Cent	158 1	130 F	110 1
10		Aropol 7010	75	1	•	9		
13   13   14   13   15   15   15   15   15   15   15	•		32.	Ashland Proprietary Inhib	0.01	ឌ		
We have a comparable   We have a comparable	<b>5</b> 0		9	*	•	<b>-</b> ≢	13	87 2
Aropol 7420 M 75	22		9	Copper maphthanate	0.5	ជ	27,	<b>.</b>
Autopol 7420 M 75	767		9	Perric naphthanate	0.5	† <sup>°</sup>	13,	, 3
Mark	₹ 6		9	Adogen 464	0.5	50,		<b>3</b> 4
Mail and Proprietary Inhib   0.50   16   55	ر ا		9	DAGA-HCI	0.75	<b>%</b>		63+ Broken
Aropol 7410 M 70 Miland Proprietary Inhib 0.01 15 19 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	<b>`</b>		9	DMLA-HC1	0.50	91	55,	æ.
Aropol 7110 75 Ashland Proprietary Inhib 0.01 15 19 1  Aropol 7110 75 Ashland Proprietary Inhib 0.02 11 28 14 13 14 13 15 15 15 15 15 15 15 15 15 15 15 15 15	8 2		9	DAGA-HCL	0.25		55,	63
Arrigol 7110   75   Action Proprietary Inhib   0.01   15   19   1	, o		OF.	;	•	<b>_</b>	18	ส
Aropol 7110 75 Ashland Proprietary Inhib 0.01 12 28 1 1 28 1 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 1 28 1 28 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 1 29 1 20 1 20	<b>,</b> 0		\ <del>&amp;</del>	Ashland Proprietary Inhib	0.01	<b>:</b>		
Aropol 7110 75 Ashland Proprietary Inkth 0.01 12 28 14 13 14 15 14 15 15 15 15 15 15 15 15 15 15 15 15 15	£ 82		, <u>P</u>	DAGA-NCL	0.25		61	130 87
Acropol 7420 M 75 Acropol 750 Morgan 464	3 8	Offic forms	75	;	•	<b>.</b>		
Aropol 7420 M 75 Mogen 464	និង		75	Ashland Promistary Inhib	0.01	ដ		
Aropol 7420 M 750 Ashland Proprietary Inhib 0.01 14 13  Aropol 7420 M 70 Ashland Proprietary Inhib 0.01 20 20  Aropol 7420 M 70 Ashland Proprietary Inhib 0.01 32 6  Aropol 7420 M 70 Ashland Proprietary Inhib 0.01 32 6  Aropol 7420 50 Ashland Proprietary Inhib 0.01 20 18  Aropol 7420 6 40 Adogen 464 11	<b>4</b> 9		<u>.</u> %	Adogen 461	0.5	Ħ	<b>%</b>	25
Aropol 7240 MC 50 Ashland Proprietary Inhib 0.01 14 13  Aropol 7240 MC 50 Ashland Proprietary Inhib 0.01 20 20 20 20 20 20 20 20 20 20 20 20 20	່ເລ		54	* *		<b>.</b>		1
35   Ashland Proprietary Inhib   14   14     30	77		9	:	•	. <del></del>	<u></u>	84
Aropol 74:0 MC 50 Ashland Proprietary Inhib 0.01 14 4 6 11 6 6 11 6 6 11 6 6 11 6 6 11 6 6 11 6 6 6 11 6	8		35	:	•	. <del></del>		
Aropol 7240 MC   50   Ashland Proprietary Inhib   0.01   20	<b>%</b>		35	Ashland Proprietary Inhib	0.01	<b>7</b> 1		
Aropol 7240 MC 50 Ashland Proprietary Inhib 0.01 20 4  Aropol 7410 M 70 Ashland Proprietary Inhib 0.01 32  Aropol 7420 50 Ashland Proprietary Inhib 0.01 20 18  Aropol 7420 50 Adogen 464 - 65 11	2		30	1 5	•	<b>.</b>		
Aropol 7420 Marol 750 Marol Proprietary Inhib 0.01 20  Aropol 7410 M 70 Ashland Proprietary Inhib 0.01 32  Aropol 7420 50 Ashland Proprietary Inhib 0.01 26  Aropol 7420 50 Adogen 464 - 65 18  Aropol 7510 M 75 - 11	<del>, ;</del>	Aronal 7240 MC	. R	:	•	9	#	ୡ
Aropol 74.10 M 70 Ashland Proprietary Inhib 0.01 16  Aropol 74.20 Ashland Proprietary Inhib 0.01 32  Aropol 74.20 50 Adogen 464 - 65  Aropol 7510 M 75 - 11	, <b>≱</b>		20	Ashland Proprietary Inhib	0.01	ର -		
Aropol 7410 M 70 Ashland Proprietary Inhib 0.01 32 6 6 18 6 40 Adogen 464 - 6.5 11 16	<b>8</b>			The state of the s	, 6	<b>+</b> ½		
Aropol 7420 Ashland Proprietary Inhib 0.01 32 6 30 Ashland Proprietary Inhib 0.01 20 18 4ropol 7420 50 Adogen 464 - 65 118 4ropol 7510 M 75 - 11	35A		60	CHARLE CIMAL MOTOR THE TITLE	•	, so		
30 Ashland Proprietary Inhib 0.01 20 18 Aropol 7420 50 Adogen 464 6 18 Aropol 7510 M 75 11	ዪ፟፟	Work) rodory	2.5	Ashland Proprietary Inhib	0.01	, sx		
Aropol 7420 50 Ashland Proprietary Inhib 0.01 20 18 - 6 18 - 6 18 10 1510 M 75 11 16	ž.		<u> </u>	•		9		
Aropol 7420 50 - 0 10 40 Adogen 464 0.5 11 Aropol 7510 M 75 11	37A		s ጽ	Ashland Proprietary Inhib	0.01	ୡୢ	ç	;
Aropol 7510 M 75 11	77	Aropol 7420	05.	• • • • • • • • • • • • • • • • • • • •	' c	0	9 7	6
Aropol 7510 M 75	<b>3</b> 2		07	Adogen 404	<·0	;	9	
	æ	Aropol 7510 M	75	•	•	#		

Per cent plastic in reain system, balance is styrene monomer.

Inhibitor used in addition to 100 ppm lydroquimone: Copper maphthanate - 8% copper maphthanate; ferric maphthanate - 6% ferric maphthanate; Adogen 46% - quaternary ammonium chloride; IMGA-HCI - dimethyl lauryl amine hydrochloride; BHA - butylated hydroxy anisole.

Polystyrene formed on inside cover of container.

Table 5 (concluded)

	Inasturated Pol	Polyester Polyester	Inhibitor		Stabilit	y - Days t	o gel et
Ref		Solids, sa	Type	Per Cent	1 851	158 F 130 F II	91
<b>38</b>	Ammol 7510 M	ž,	Ashland Promitetary Inhib	0.01	7,1		
į Š&	E OTO TOTO TO	3		•	ľ	ส	25
2		<u> </u>	Adogen 464	0.5	<b>S</b> 2	150	
ori Ori		3	DOA-HCL	0.1	5	101	
יות		9	DGA-HC1	0.75	સ	#	
व		O <del>t</del>	IMIA-HC1	5.0	\$	130	
9T		<b>Q</b>	DMA-RCI	<b>1</b> 0	;	55	
84	Aropol 7510 M	9	HILY HILL HILL HILL HILL HILL HILL HILL	70.0	æ:	;	
113		ጸ	BILA	1.0	;	23	
זָר דו		ጸ	BHA	0.25	<b>≓</b> :	χ:	
115		æ	BHA	0.5	라 :	Σ.;	
8,		E	Adogen 464	0.25	15	115	
95		<u>ب</u>	Adogen 464	5.0	13	٤.	
ま		R	Adogen 464	1.0		Ž,	
26		S. S.	Adogen 464/Cu Raph	0.4/0.01	ድ	190	
· &		30	DALA-NCI	0.5	<b>*</b>	ເຊິ່	
8		R	DAZA-HCI	0.25	57 ;	8	
27	Aropol 7720 M	ક્	:	•	TT 1		
₹ş	•	65	Ashland Proprietary Inhib	10.0	₹ '	;	
87		01	;	,	o,	<b>≓</b>	
8		3	Adogen 464	5.0	ជ	<del>1</del> 2	₩.
194	Aronol o6140	02	:	1	ส		
16A		20	Ashland Proprietary Inhib	0.01	8 <u>7</u> `		
17.7		8	Ashland Proprietary Inhib	10.0	97	;	i
	Aropol 7010/7510 M	<u>9</u>	Adogen 464	V. 0		£ d	ድ ኢ
# d	Aropol 7420/7510 M		Adogen 404			7 7	, G
<b>8</b> 5	Aropol 7510 M/7720 M		Adogen 404			ţ	5

Tensile Strength Ultimate (100' Elon- 300' Elon- Zionpsiton gelion, psi gation, psi Per Cent										``	20	523		635	570	3.	790		2307		₹.	ş					165	133	28	,140	157	187	275		
	026	750	¥	345	01.4		510	;	Ŕ	:	<u>ጸ</u>	9 1	Cot T	N75	o).	;	2	8 }	107	₹,	ફ	01,	217	į	9 5	) e	21	æ	13	<b>%</b>	8	3	830	2	ì
Elongation,							1125							4.75																					
Teneille Strength Poi										_																									
Catalyst		DARCO	DABCO	DALBCO	DABCO	Co Map	DANCO	8	DAMES.	2 2 3	8	2		100TOC	Darte		TENT		DETEC	DECTO	USTOL	Ì				Decor	DETTE	H	DEFEDE		DELDE		DEFERE	7000	
Ratio																												7/52							
Curing Agent		Mofetume	Motature	Moteture	Moisture		Motsture		Mo1 sture	Triol	Polyester diol	Moisture	GLy cert ne	Polyester Diol	Aron din-	Polyether Diol	425 MV DIOL	Die 200-Diol 134	DIA 200-Diol 52	Die 200-Die1 300	Die 200-mon 425	1/2 theo use level	Me 200-Mot 72	20 Mar 200 Mar 1,056	Me 200-1501 134	We soo-med Jak	Die 200-Diol 52	Ma 200-Mol 52	Die 200-Diol 52	PLB 200-PHOL 90	Dia 200-Diol 52	Dia 200-Diol 52	Ma 200-Diol 90	Dia 200-Diol 52	
NCO/OH PAtio		4 5/7	3.5/1	3.5/1	3.5/1		3/1		3/1	3,1	3.5/1	3.5/1	4/1	3.5/1	3/1/		2/1	2/1	2/2	₹/e	2/1	;	7/2	1,0	100	170	5/2	2/2	2/1	۲/2	2/1	2/1	2/7	7.3	
Restn	torab Tates, 121	MAN LANCEX 131	707										Polyenter Amide	Polvester Mol Th	Polyether Mol Th		2000 NA DE01-HDI																		
Run No.		or all direct	(5-(0-0-(3-)	1	i d		1.4		<b>6</b> 9	G	52	<b>&amp;</b>	यू	287	8		Ş,	91.7		. 25 26	<u></u>	,	<b>8</b>	<b>3</b>	21.3	X.S	375	28.	<b>3</b> 5	ę S	ું જ	9	9	ĝ	3

a. Refers to nolar ratio of Isocyanate to hydroxyl in prepolymer.
 b. Numbers refer to nolecular weight. Ratio refers to molar ratio of dismin, to diol.
 c. Catalyst used to promote cure: DESCO - tertiary and mains; Co Naph - 6% cobalt maphthenate; Sn Octo - 8% Stannous octoate; and DETDL - dibutyltin dilaurate.
 c. Elongation beyond maximum of machine, did not break.
 e. 65% in NeCl2

Table 7 Date Used in Evaluation of Cosine 0

	Surface Tension of Liquid Relative						
Soil	Liquid	dynes/cm	to H2O	Slope	Slope		
Sand	EtOH H <sub>2</sub> O 2M CaCL <sub>2</sub> 4M CaCl <sub>2</sub> 5M CaCl <sub>2</sub>	22.27 72.3 78.9 86.9 90.4	1.19* 11.0 * 1.85** 4.4** 8.5**	1.80 <del>7</del> <del>1</del> 1.33 1.029	1.0 7 1 0.519 0.577		
Silt	EtOH H <sub>2</sub> O 2M CaCl <sub>2</sub> 4M CaCl <sub>2</sub> 5M CaCl <sub>2</sub>	22.27 72.3 78.9 86.9 90.4	1.19* 11.0 * 1.85** 4.4** 8.5**	0.592 0.500 0.432 0.267 0.125	1.0 0.178 0.231 0.193 0.0786		
Clay	EtOH H <sub>2</sub> O 2M CaCl <sub>2</sub> 4M CaCl <sub>2</sub> 5M CaCl <sub>2</sub>	22.27 72.3 78.9 86.9 90.4	1.19* 11.0 * 1.85** 4.4** 8.5**	0.35 7 0.178 0.167	1.0 7 0.372 0.393		

Data from Lang's Handbook. Data from Internation Critical Tables Not calculated - see text.

Table 8 Prediction of Spreading of Resin Systems on Soil Samples

	Spreading Coefficient  yL S(equation 12)*  dynes/ dynes/cm			Adhesion			
Resin System	cm	Sand	Clay	Silt	Sand	Clay	Silt
Aropol 7110 Styrene	45 <b>3</b> 2	+36	+28	+17	Good	Good	Fair
Arothane 170 C <sub>18</sub> NCO	29 <b>3</b> 1	+37	+29	+18	Good	Good	Fair
Chem Rez A-200	41	+27	+19	+ 8	Good	Fair	Fair
Sodium Silicate "N"	<b>7</b> 7	-19	-17	-28	Poor	Poor	Poor

Note: \* $\gamma_S$  values are correlated with lowest  $\gamma_T$  value of resin systems since components with low  $\gamma_L$  tend to be surface active.

 $<sup>\</sup>gamma_{\rm S}$  (Sand) = 69 dynes/cm,  $\gamma_{\rm S}$  (Clay) = 60 dynes/cm,  $\gamma_{\rm S}$  (Loam) = 49 dynes/

## Security Classification

DOCUMENT CONTRO	OL DATA - R & D	
curity classification of title, body of abetract and indexing an	notation must be entered when the overall repor	t is classified)
ING ACTIVITY (Corporate author)	20. REPORT SECURITY CL	
Chamilan A Ganasa		

Ashland Chemical Company Minneapolis, Minn.

Unclassified

A REPORT TITLE

RESEARCH STUDY ON SOIL TREATMENT MATERIALS FOR DUST PALLIATION, SOILS WATERPROOFING AND SOIL STRENGTHENING

4. DESCRIPTIVE NOTES (Type of report and inclusive dates)

Final report

(50

8. AUTHOR(5) (First name, middle initial, last name)

C. N. Impola

D. A. Olsen

. REPORT DATE	78. TOTAL NO. OF PAGES 75. NO. OF REFS			
November 1968	45			
M. CONTRACT OR GRANT NO.	M. ORIGINATOR'S REPORT NUMBER(S)			
DA-22-079-eng-437				
S. PROJECT NO.				
1-T-0-62103-A-046-05				
c.	Sh. OTHER REPORT NO(5) (Any other numbers that may be assigned			
	U. S. Army Engineer Waterways Experiment			
4	Station, Contract Report S-68-5			

10. DISTRIBUTION STATEMENT

Each transmittal of this document outside the agencies of the U. S. Government must have prior approval of U. S. Army Materiel Command.

11. SUPPLEMENTARY NOTES

Prepared under contract for U. S. Army Engineer Waterways Experiment Station, Vicksburg, Miss.

12. SPONSORING MILITARY ACTIVITY

U. S. Army Materiel Command

2. Additional The object of this study was to evaluate various commercially-available resin systems, both organic and inorganic, to determine if they could be utilized as dust palliatives, soil waterproofing agents, and soil stabilizers. The screening tests revealed that many of the resin systems would perform as dust palliatives or soil waterproofing agents for nontraffic areas. However, none of these used on loose soil at the low use level of 3 lb/sq yd had enough strength or flexibility to perform satisfactorily under traffic. At two or three times this rate (6-9 lbs/sq yd), sufficient strength could be obtained with epoxy and unsaturated polyester resins to withstand some traffic. The emphasis was then changed to laboratory-synthesized urethane elastomers with over 2000% elongation and over 1000 psi tensile strength. When applied on loose sand, these elastomers gave a tough, flexible surface coating. Some problems with surface cracking were encountered on loose, dry clay, due to shrinkage. However, it was possible to eliminate or minimize these cracks by formulation changes of the elastomer and its curing agent, and by a very light prewetting of the soil with water. The discussion of the test results in the main body of the report is in chronological order. The Appendix covers the work done under the Physical Chemistry studies early in the contract period.

Unclassified

## Unclassified

Security Classification		LINK A	LINK Danne	E & LINK C	
ARY WORDS !		ROLE SEWY	ROLE - WTG	ROLE	WT
Dust control			<b>***</b>		***
Elastomers					
Epoxy resins					
Polyester resins					
Resins					
Soil stabilization					
Soil waterproofing					
Waterproofing					
					1 (A)
					8
					177
				44.4	
	ing Arts San San				
		The second of th	la.		
	23 - 19 19 3				
	and the second				
				. .	

Inclassified

Secretary Constitution